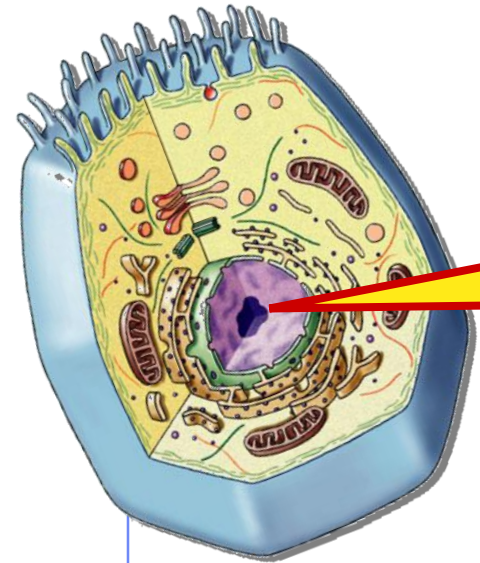


Cellular Respiration

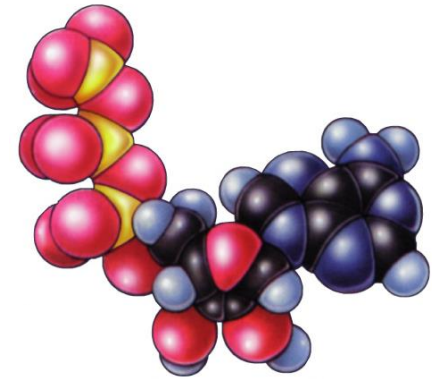
Harvesting Chemical Energy

ATP



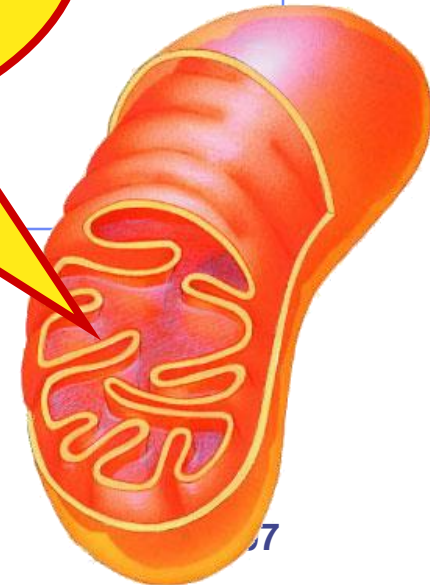


What's the point?



The point is to make **ATP!**

ATP



Harvesting stored energy

- Energy is stored in organic molecules
 - ◆ carbohydrates, fats, proteins
- Heterotrophs eat these organic molecules → food
 - ◆ digest organic molecules to get...
 - raw materials for synthesis
 - fuels for energy
 - ◆ controlled release of energy
 - ◆ “burning” fuels in a series of step-by-step enzyme-controlled reactions



Harvesting stored energy

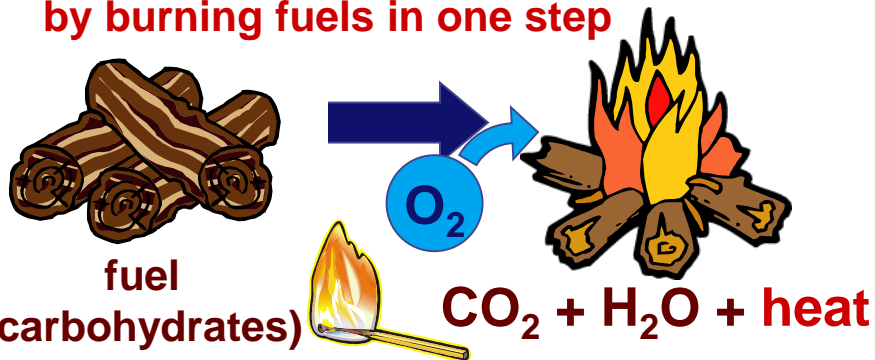
- Glucose is the model
 - catabolism of glucose to produce ATP

respiration

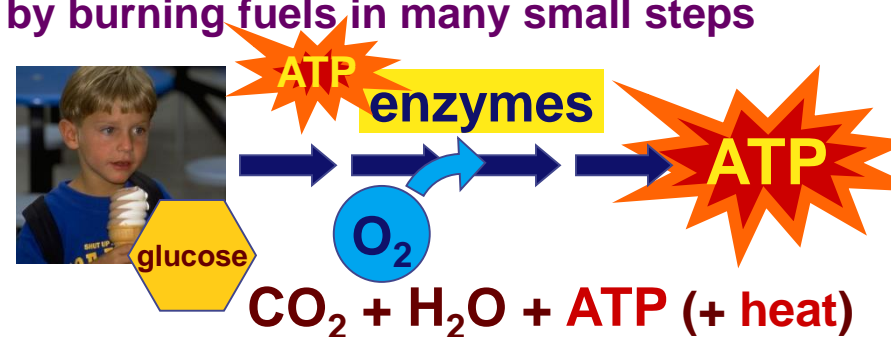
glucose + oxygen → energy + water + carbon dioxide



COMBUSTION = making a lot of heat energy by burning fuels in one step

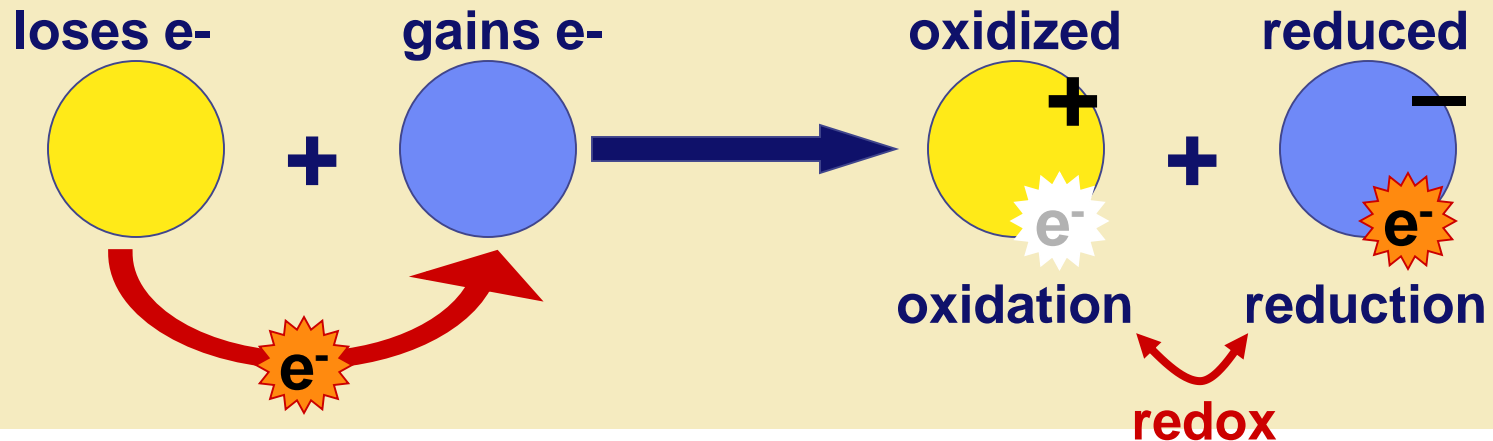


RESPIRATION = making ATP (& some heat) by burning fuels in many small steps



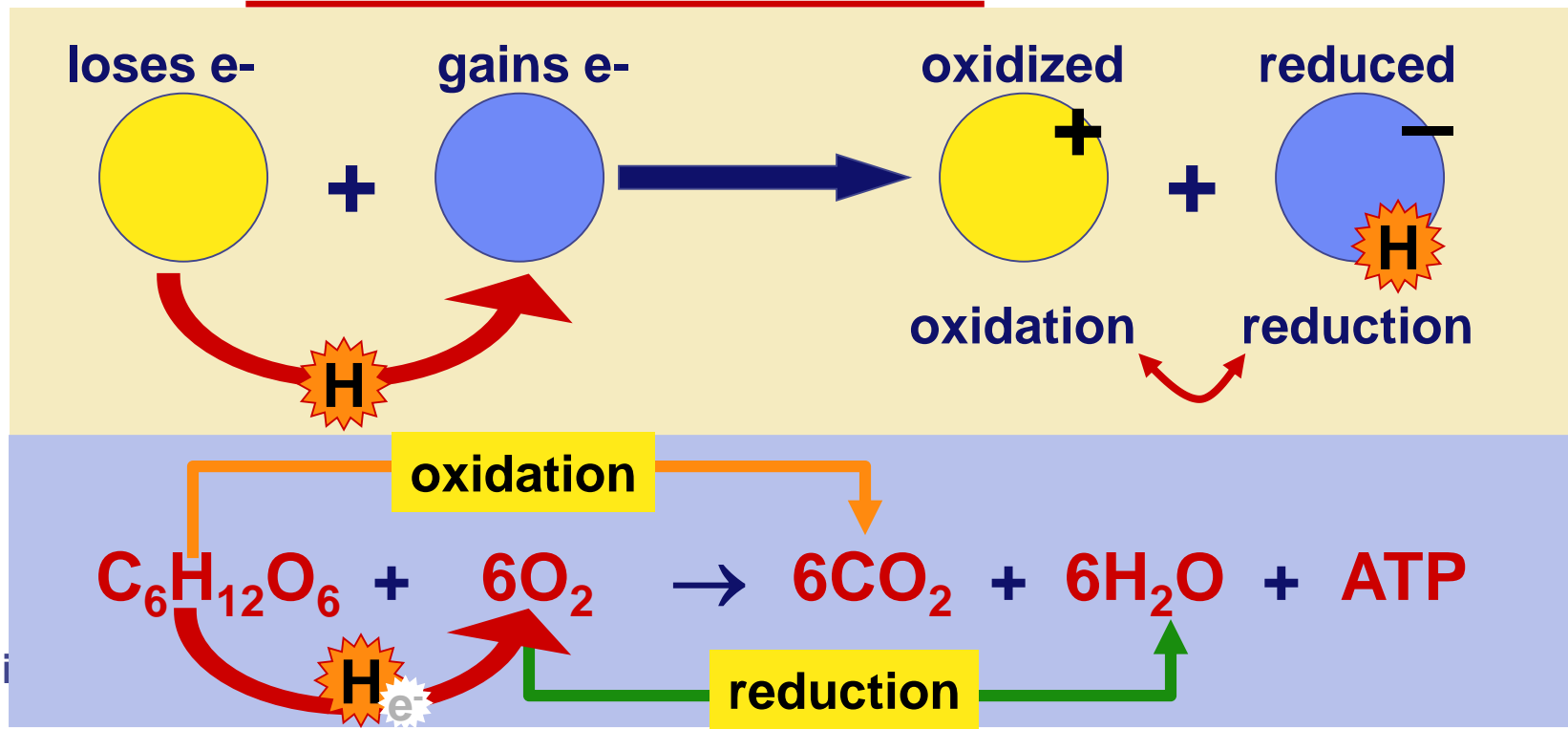
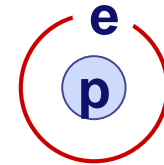
How do we harvest energy from fuels?

- Digest large molecules into smaller ones
 - ◆ break bonds & move electrons from one molecule to another
 - as electrons move they “carry energy” with them
 - that energy is stored in another bond, released as heat or harvested to make ATP




How do we move electrons in biology?

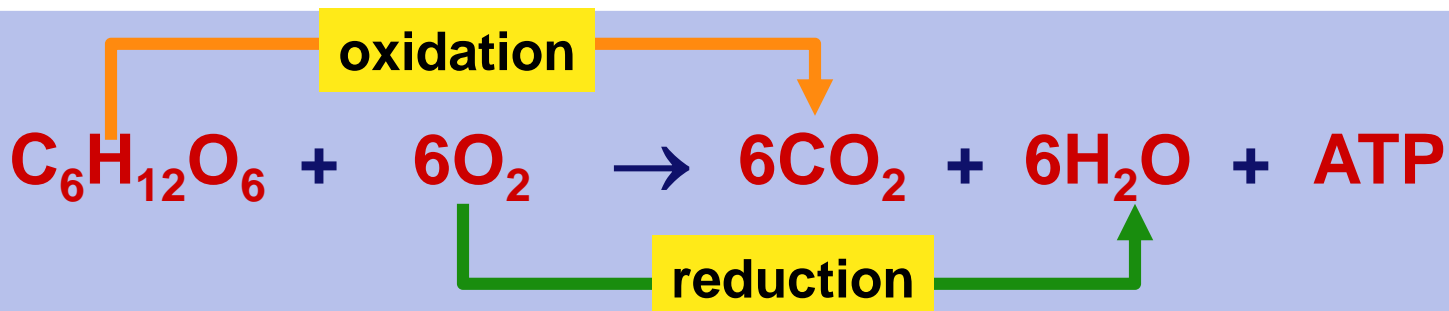
- Moving electrons in living systems
 - ◆ electrons cannot move alone in cells
 - electrons move as part of H atom
 - move H = move electrons



Coupling oxidation & reduction

REDOX reactions in respiration

- ◆ release energy as breakdown organic molecules
 - break C-C bonds
 - strip off electrons from C-H bonds by removing H atoms
 - ◆ $C_6H_{12}O_6 \rightarrow CO_2$ = the fuel has been oxidized
 - electrons attracted to more electronegative atoms
 - ◆ in biology, the most electronegative atom? 
 - ◆ $O_2 \rightarrow H_2O$ = oxygen has been reduced
- ◆ couple REDOX reactions & use the released energy to synthesize ATP²



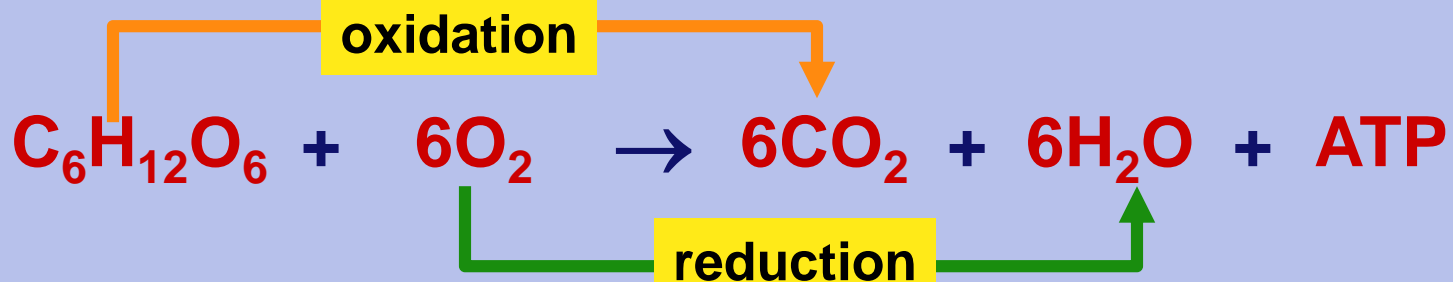
Oxidation & reduction

■ Oxidation

- ◆ adding O
- ◆ removing H
- ◆ loss of electrons
- ◆ releases energy
- ◆ exergonic

■ Reduction

- ◆ removing O
- ◆ adding H
- ◆ gain of electrons
- ◆ stores energy
- ◆ endergonic



Moving electrons in respiration

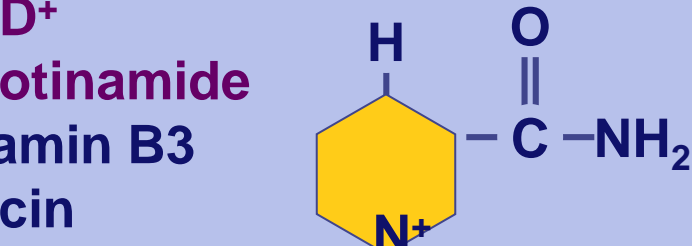
- **Electron carriers** move electrons by shuttling H atoms around
 - ◆ $\text{NAD}^+ \rightarrow \text{NADH}$ (reduced)
 - ◆ $\text{FAD}^{+2} \rightarrow \text{FADH}_2$ (reduced)

like \$\$
in the bank



reducing power!

NAD^+
nicotinamide
Vitamin B3
niacin

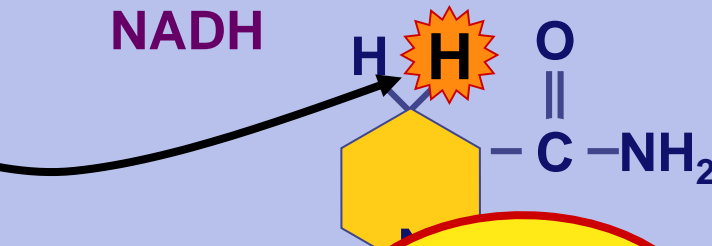


+



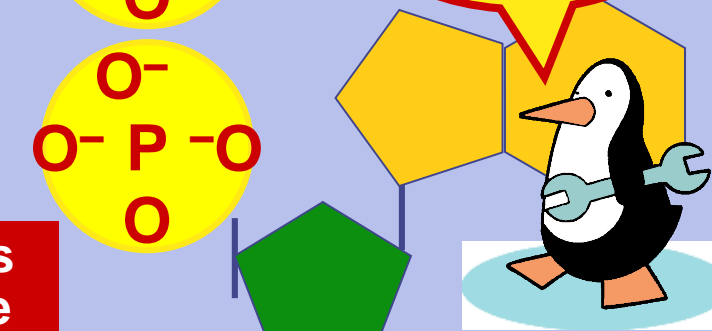
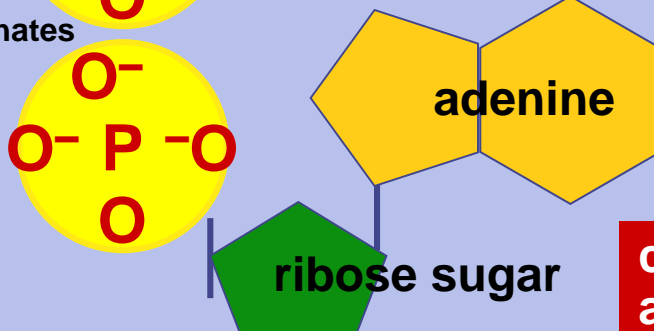
reduction
oxidation

NADH

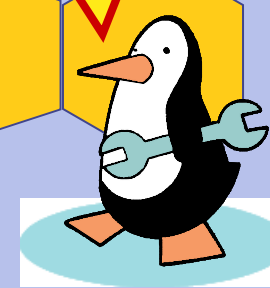


How efficient!
Build once,
use many ways

phosphates



carries electrons as
a reduced molecule



Overview of cellular respiration

■ 4 metabolic stages

◆ Anaerobic respiration

1. Glycolysis

- ◆ respiration without O_2
- ◆ in cytosol

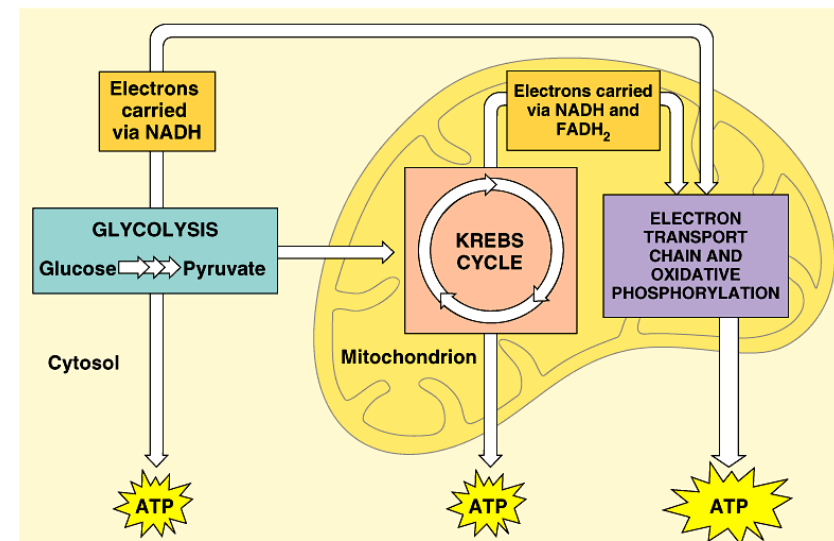
◆ Aerobic respiration

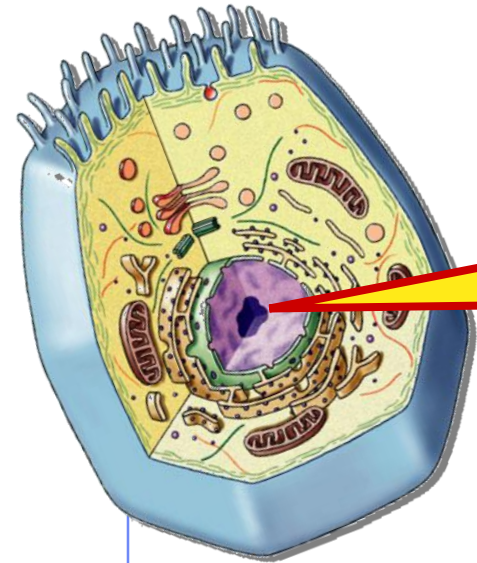
- ◆ respiration using O_2
- ◆ in mitochondria

2. Pyruvate oxidation

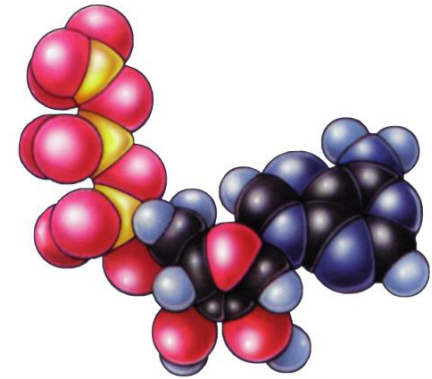
3. Krebs cycle

4. Electron transport chain



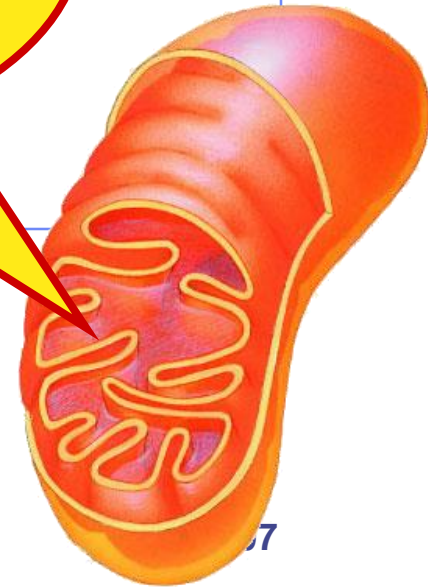


What's the point?



The point is to make **ATP!**

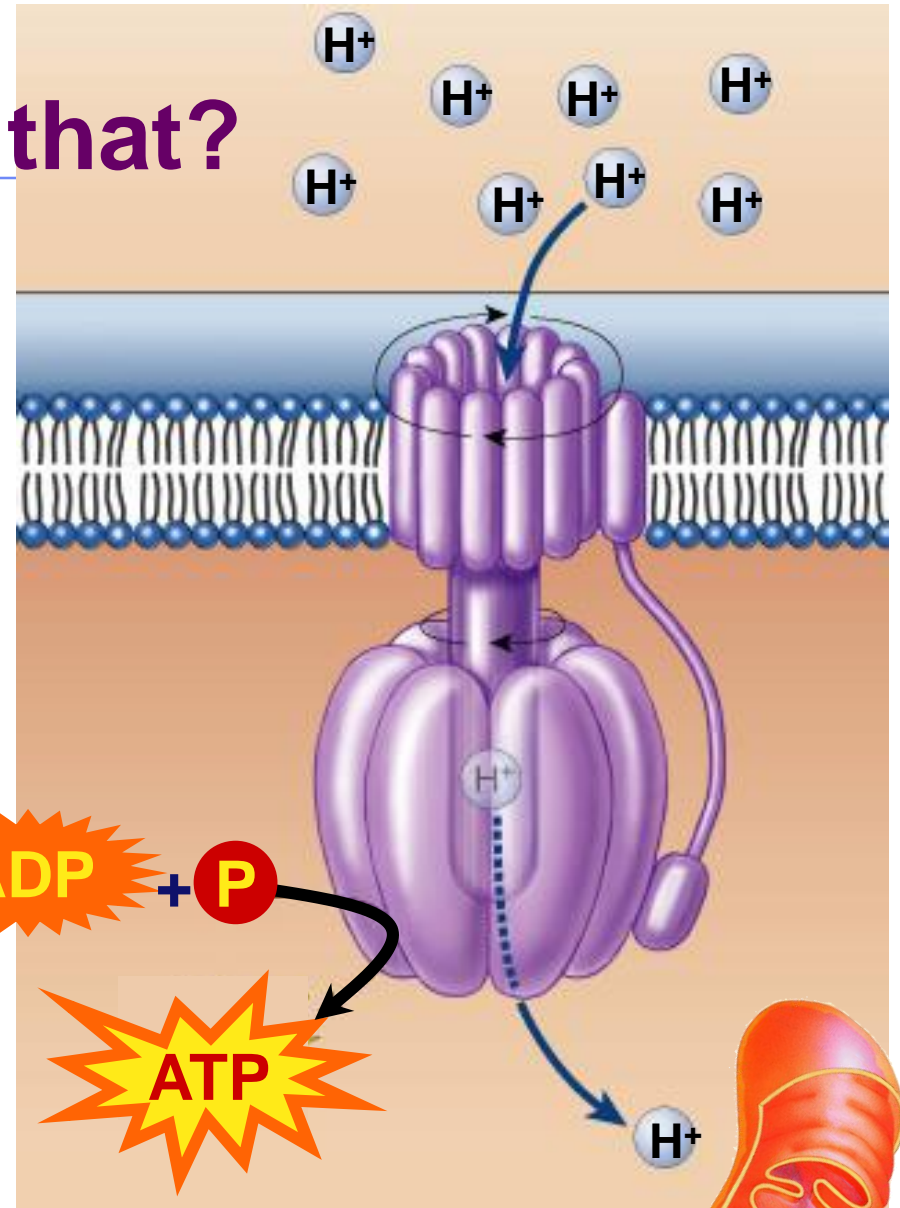
ATP



And how do we do that?

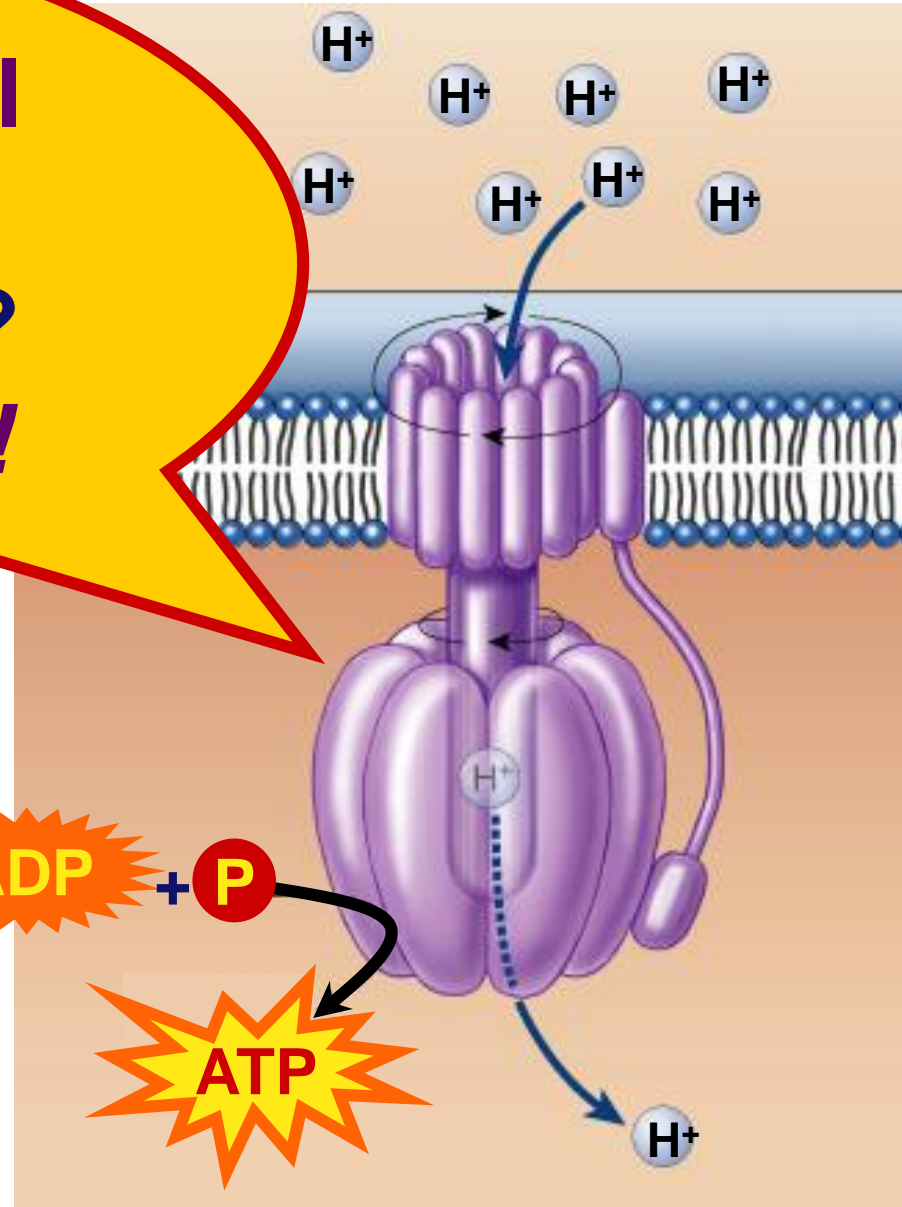
■ ATP synthase enzyme

- ◆ H^+ flows through it
 - conformational changes
 - bond P_i to **ADP** to make **ATP**
- ◆ set up a H^+ gradient
 - allow the H^+ to flow down concentration gradient through ATP synthase
 - **$ADP + P_i \rightarrow ATP$**



But... How is the proton (H^+) gradient formed?

Got to wait until
the sequel!
Got the Energy?
Ask Questions!



Cellular Respiration

Stage 2 & 3:

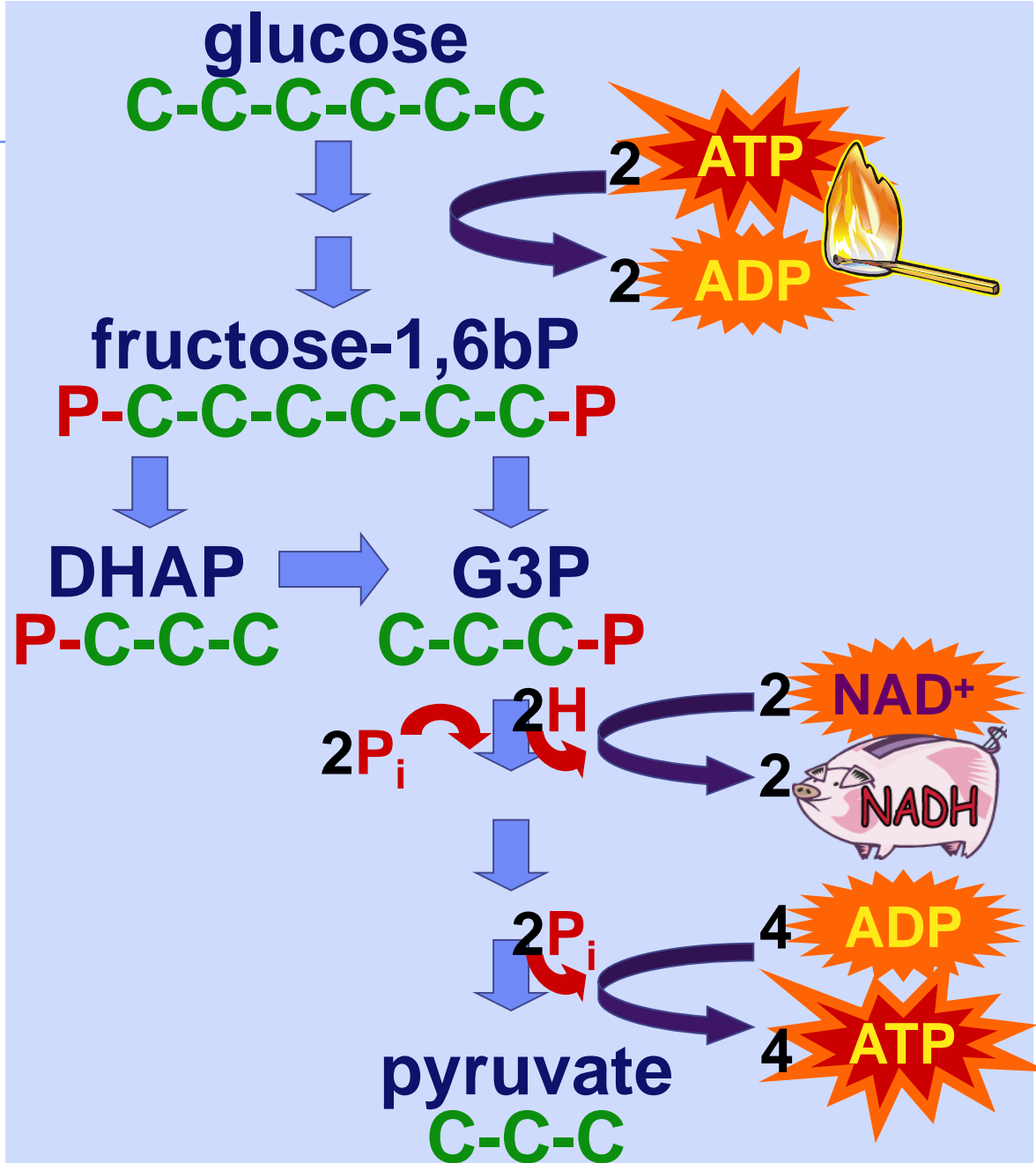
Oxidation of Pyruvate Krebs Cycle



Overview

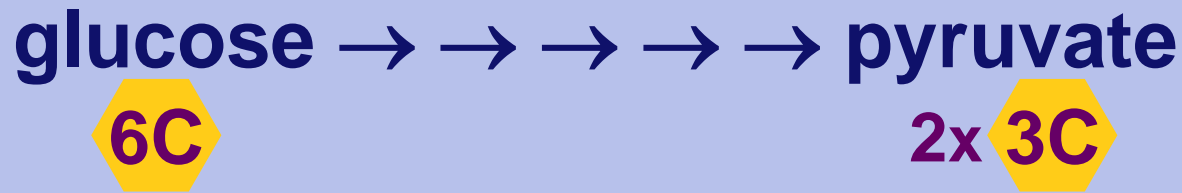
10 reactions

- ◆ convert glucose (6C) to 2 pyruvate (3C)
- ◆ produces: 4 ATP & 2 NADH
- ◆ consumes: 2 ATP
- ◆ net: 2 ATP & 2 NADH



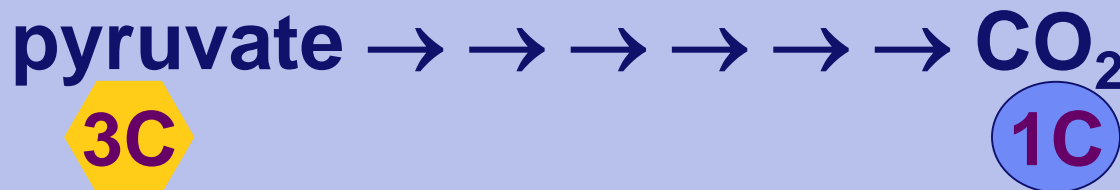
Glycolysis is only the start

- Glycolysis

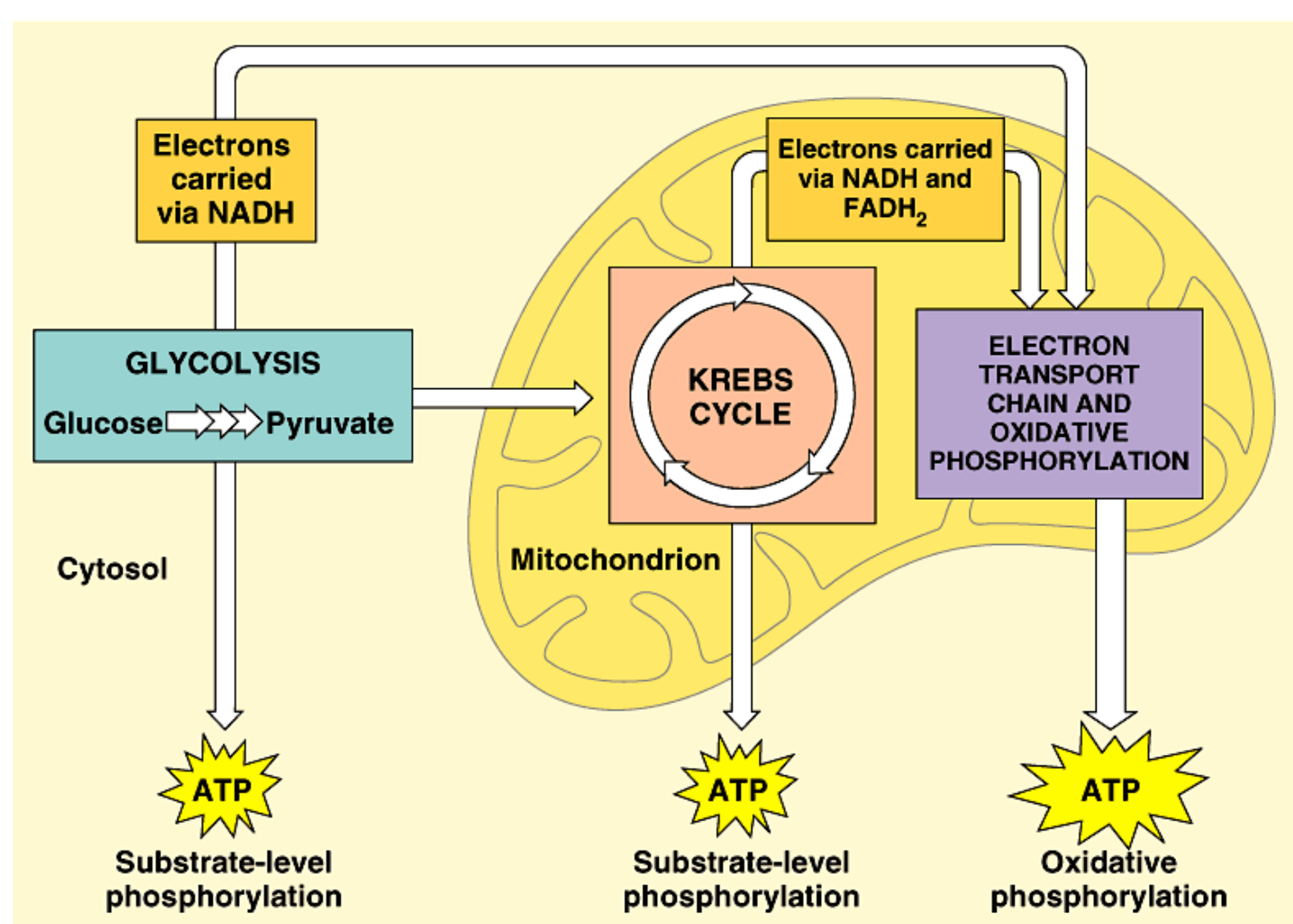


- Pyruvate has more energy to yield

- ◆ 3 more C to strip off (to oxidize)
- ◆ if O_2 is available, pyruvate enters mitochondria
- ◆ enzymes of Krebs cycle complete the full oxidation of sugar to CO_2



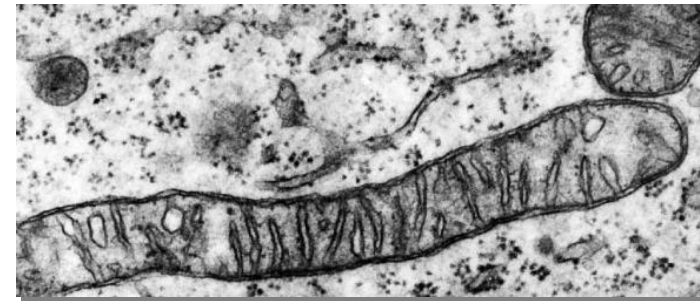
Cellular respiration



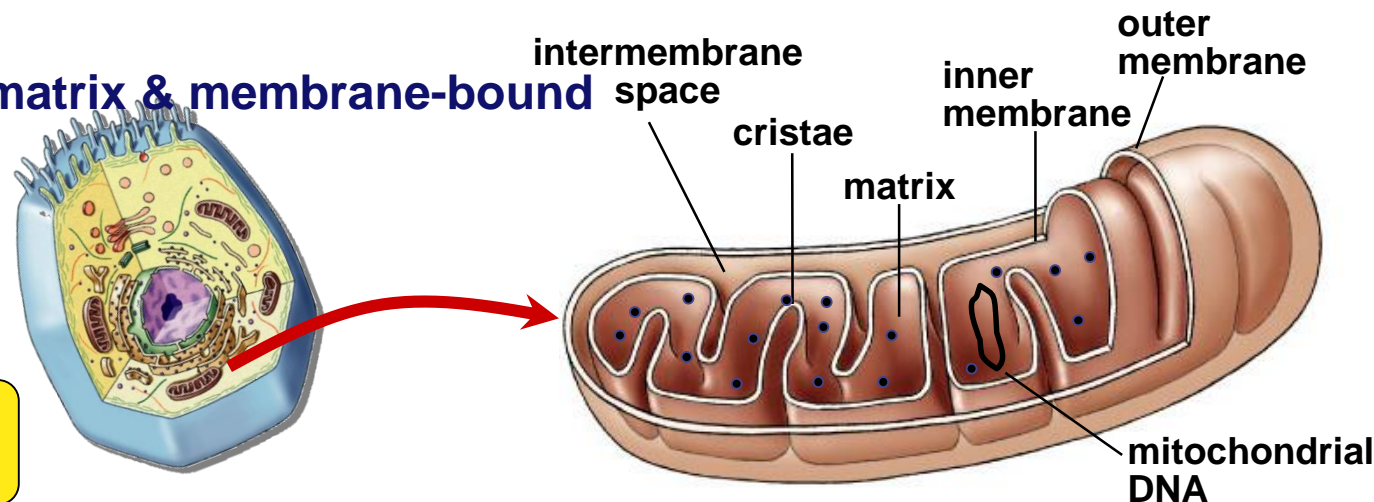
Mitochondria — Structure

- Double membrane energy harvesting organelle

- ◆ smooth outer membrane
- ◆ highly folded inner membrane
 - cristae
- ◆ intermembrane space
 - fluid-filled space between membranes
- ◆ matrix
 - inner fluid-filled space
- ◆ DNA, ribosomes
- ◆ enzymes
 - free in matrix & membrane-bound

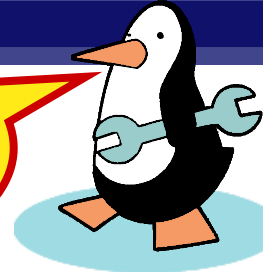


What cells would have a lot of mitochondria?



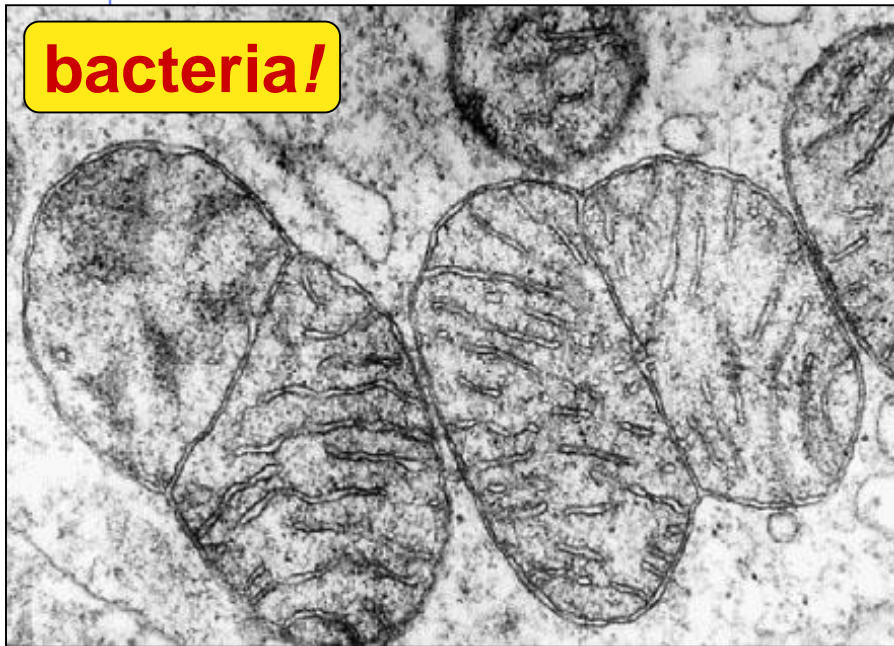
Mitochondria – Function

Oooooh!
Form fits
function!



Dividing mitochondria
Who else divides like that?

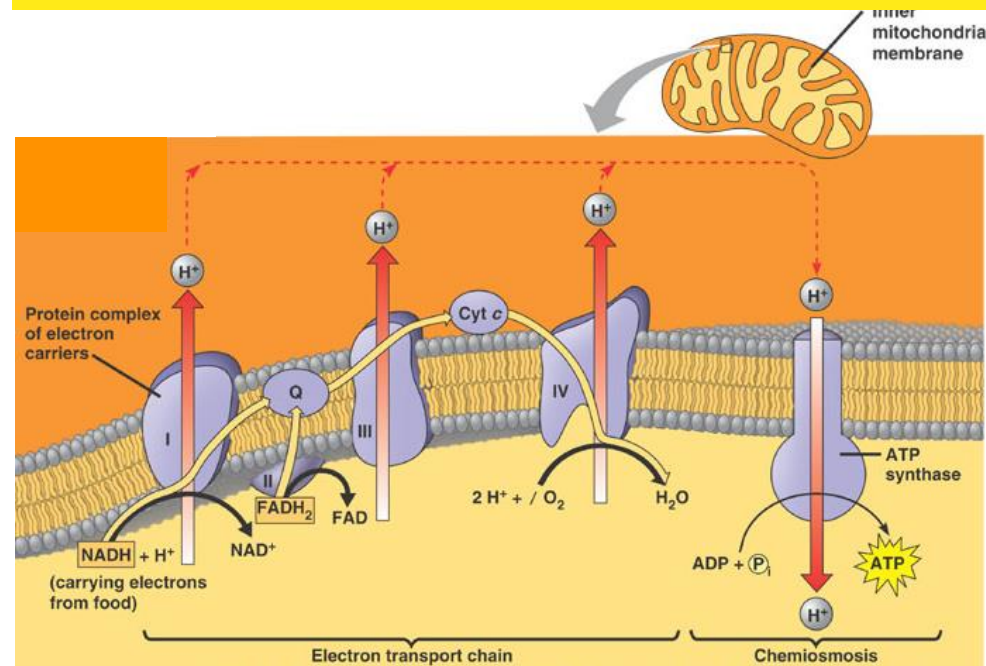
bacteria!



What does this tell us about the evolution of eukaryotes?

Endosymbiosis!

Membrane-bound proteins
Enzymes & permeases

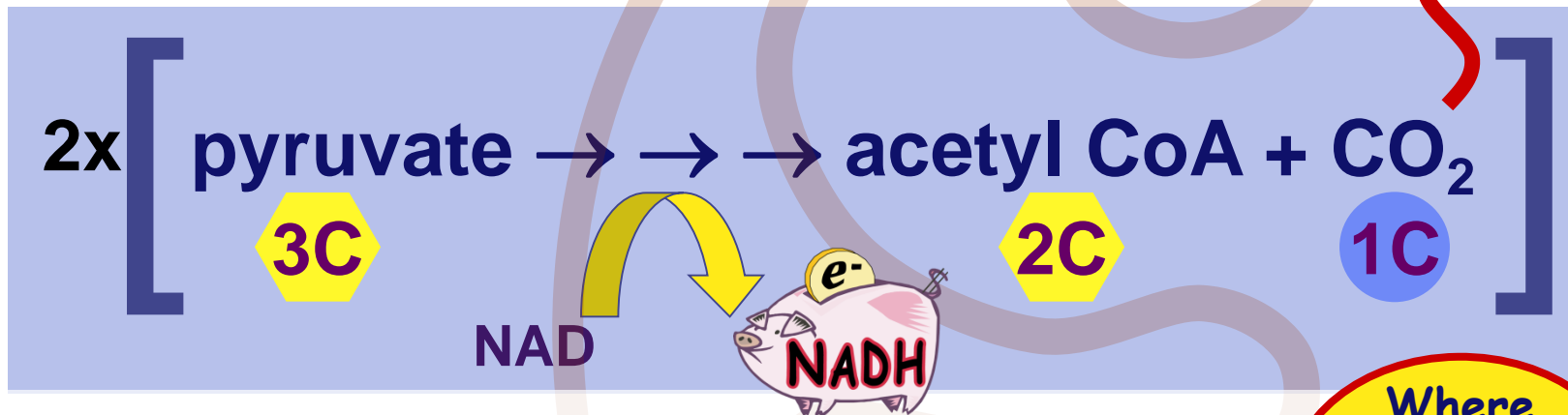


Advantage of highly folded inner membrane?

More surface area for membrane-bound enzymes & permeases

Oxidation of pyruvate

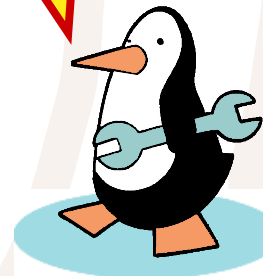
- Pyruvate enters mitochondrial matrix



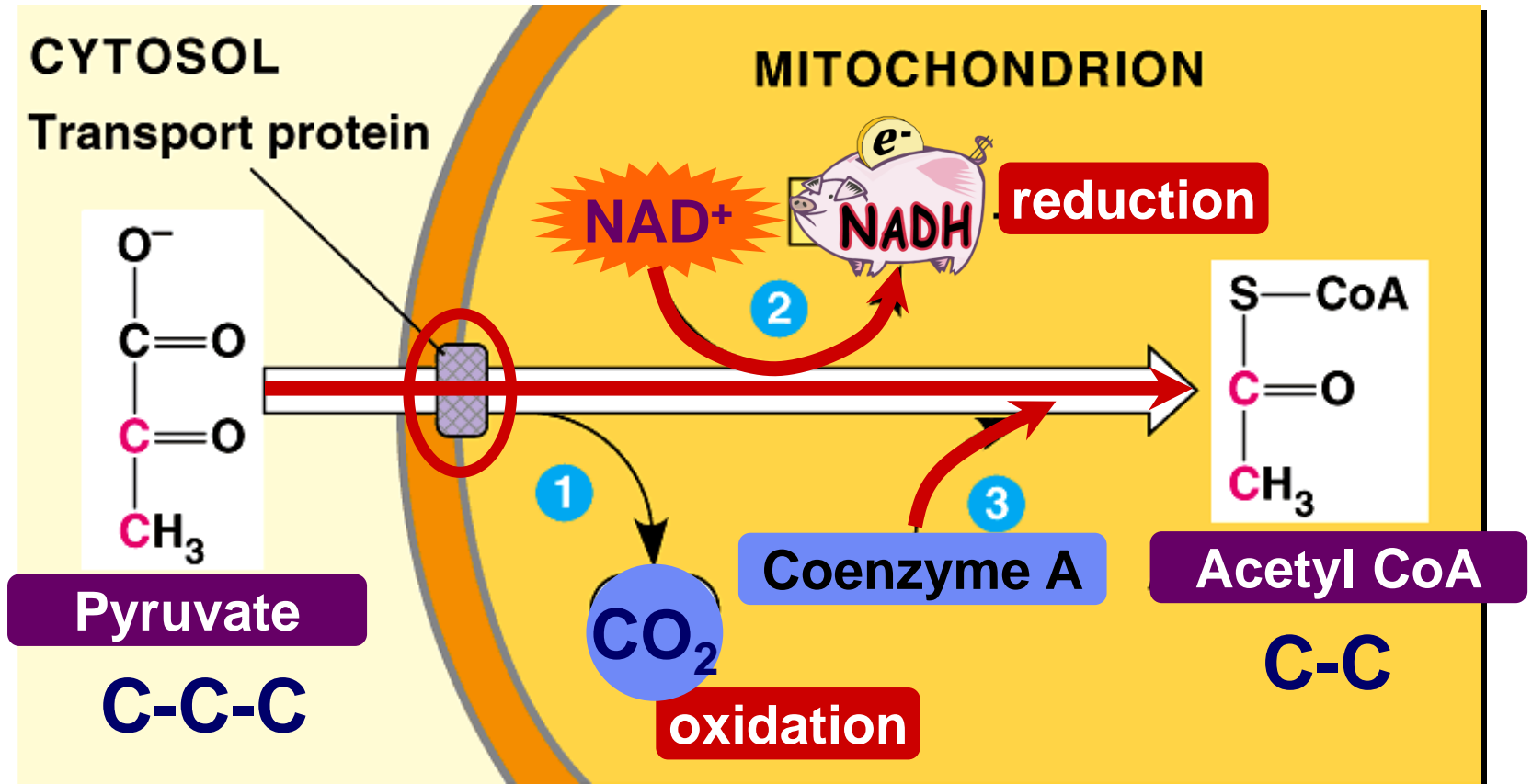
- 3 step **oxidation** process
- releases **2 CO₂** (count the carbons!)
- reduces **2 NAD → 2 NADH** (moves e⁻)
- produces **2 acetyl CoA**

- Acetyl CoA enters **Krebs cycle**

Where
does the
CO₂ go?
Exhale!



Pyruvate oxidized to Acetyl CoA



$$2 \times \left[\text{Yield} = 2\text{C sugar} + \text{NADH} + \text{CO}_2 \right]$$

1937 | 1953

Krebs cycle



Hans Krebs
1900-1981

- aka Citric Acid Cycle
 - ◆ in mitochondrial matrix
 - ◆ 8 step pathway
 - each catalyzed by specific enzyme
 - step-wise catabolism of 6C citrate molecule
- Evolved later than glycolysis
 - ◆ does that make evolutionary sense?
 - bacteria → 3.5 billion years ago (glycolysis)
 - free O₂ → 2.7 billion years ago (photosynthesis)
 - eukaryotes → 1.5 billion years ago (aerobic respiration = organelles → mitochondria)

Count the carbons!

pyruvate **3C** → **2C** acetyl CoA

4C → **6C** citrate

4C

oxidation
of sugars

6C

CO₂

4C

x2

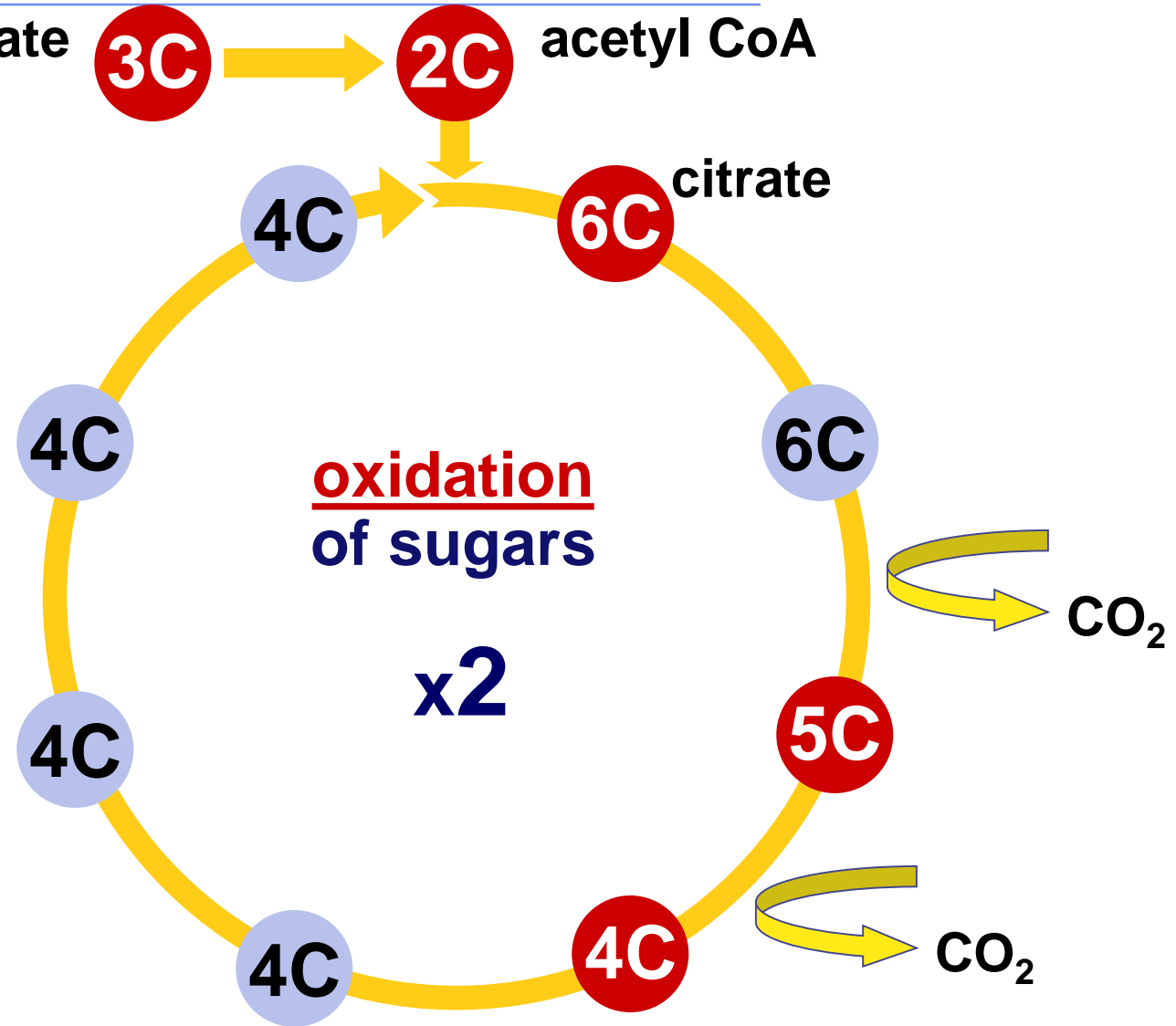
5C

CO₂

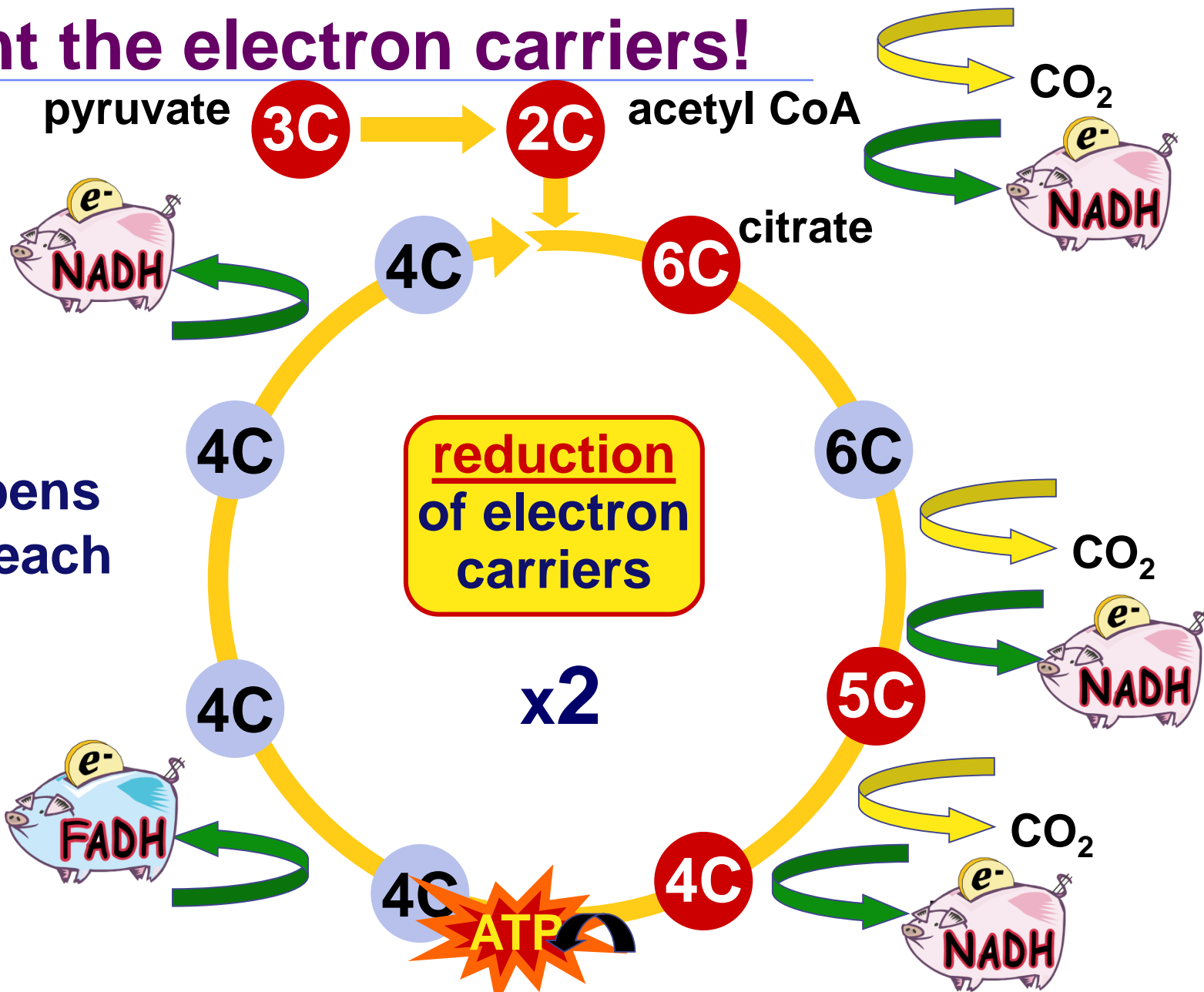
4C

4C

This happens twice for each glucose molecule



Count the electron carriers!



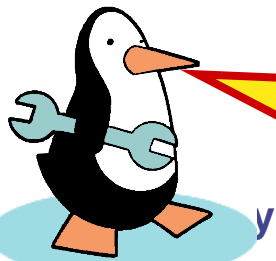
This happens twice for each glucose molecule

Whassup?

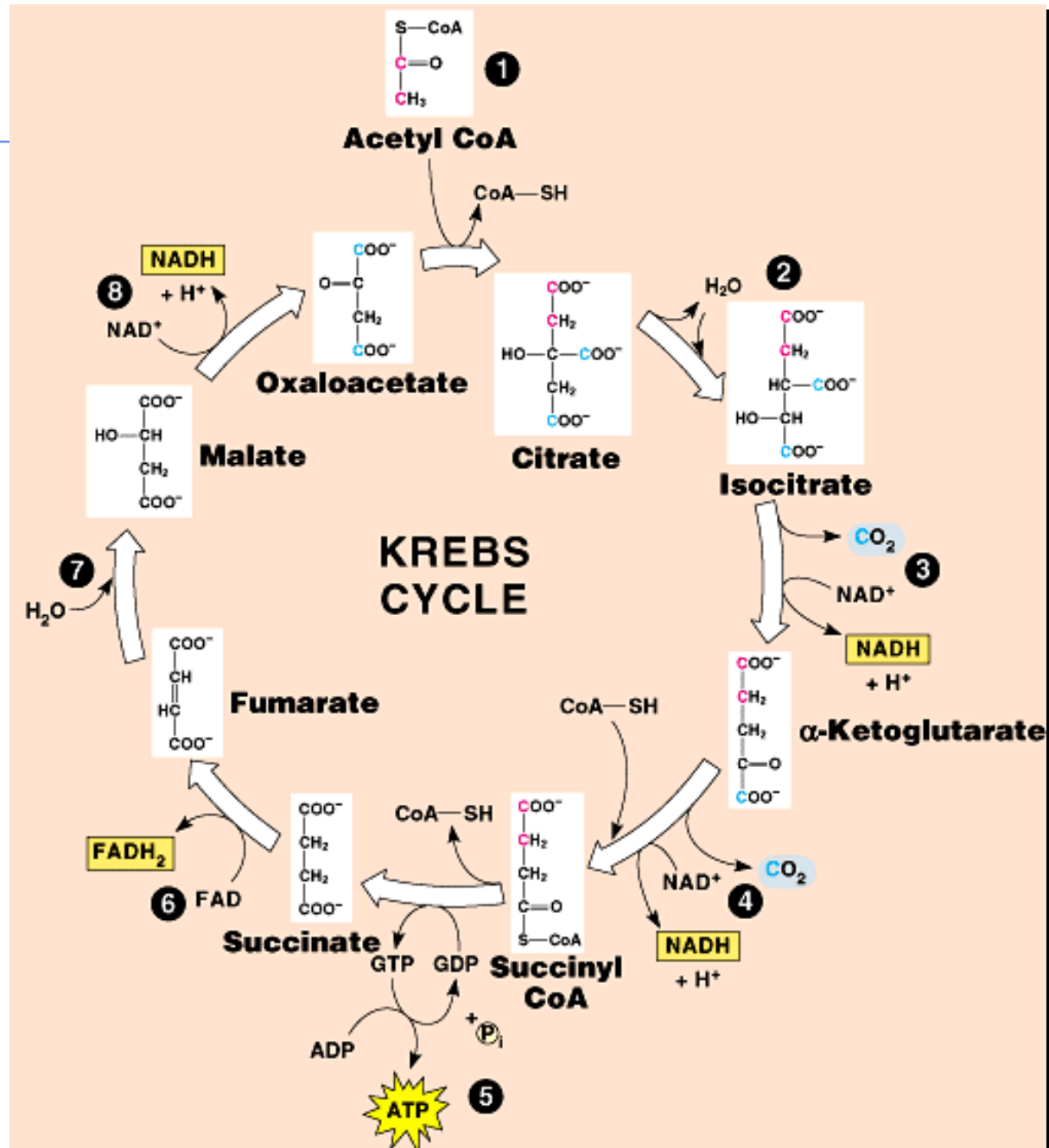
So we fully oxidized glucose



& ended up with 4 ATP!

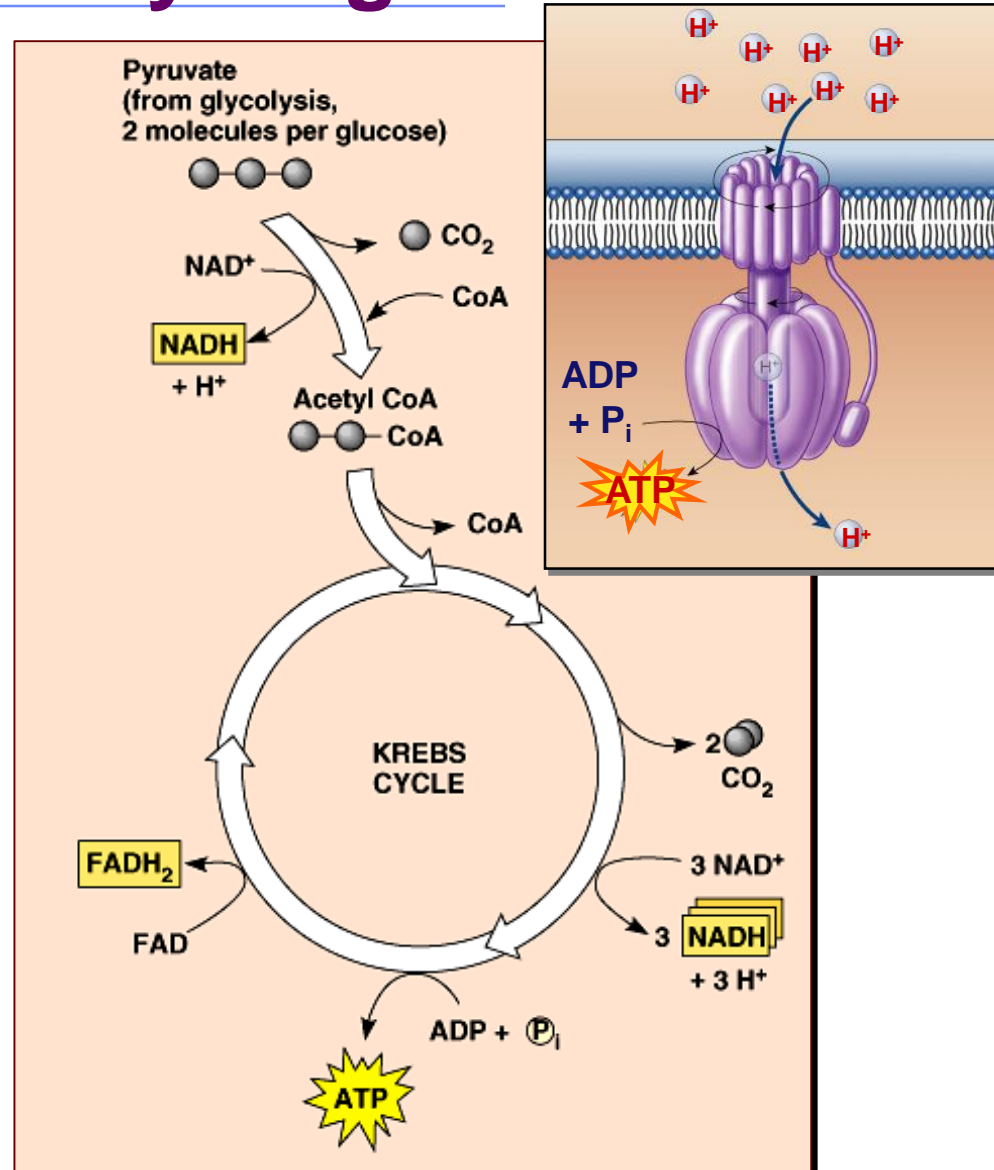


What's the point?

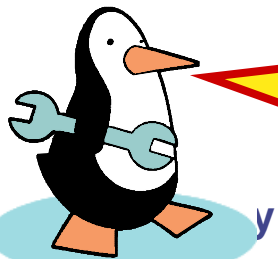


Electron Carriers = Hydrogen Carriers

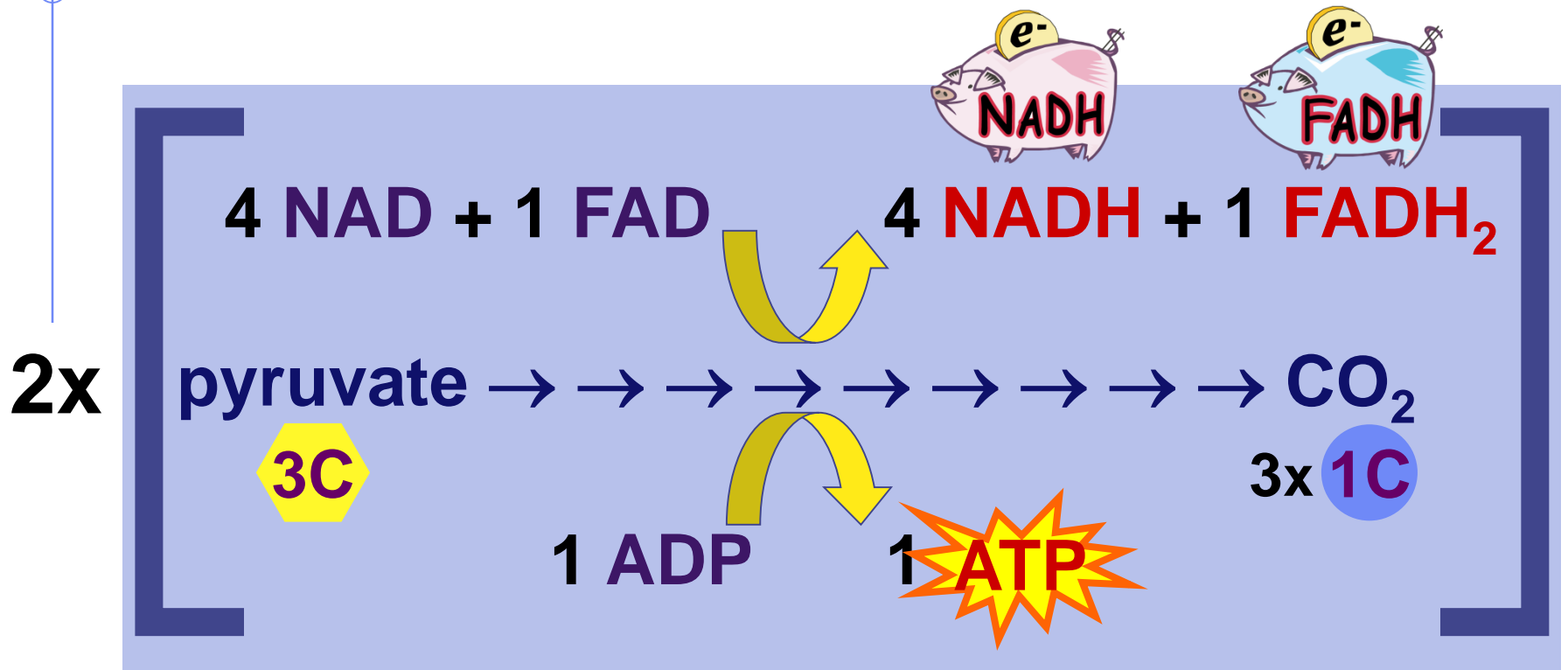
- Krebs cycle produces large quantities of electron carriers
 - ◆ NADH
 - ◆ FADH₂
 - ◆ go to Electron Transport Chain!



What's so important about electron carriers?



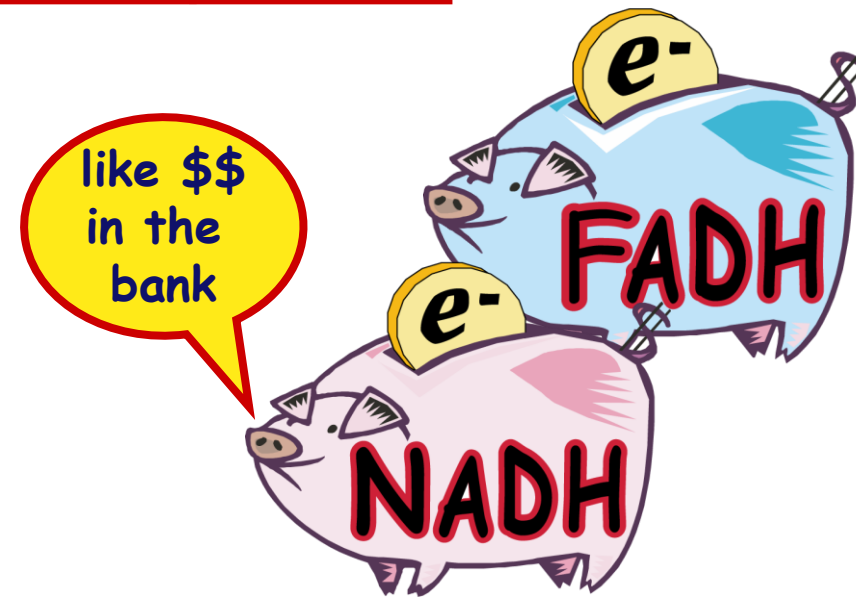
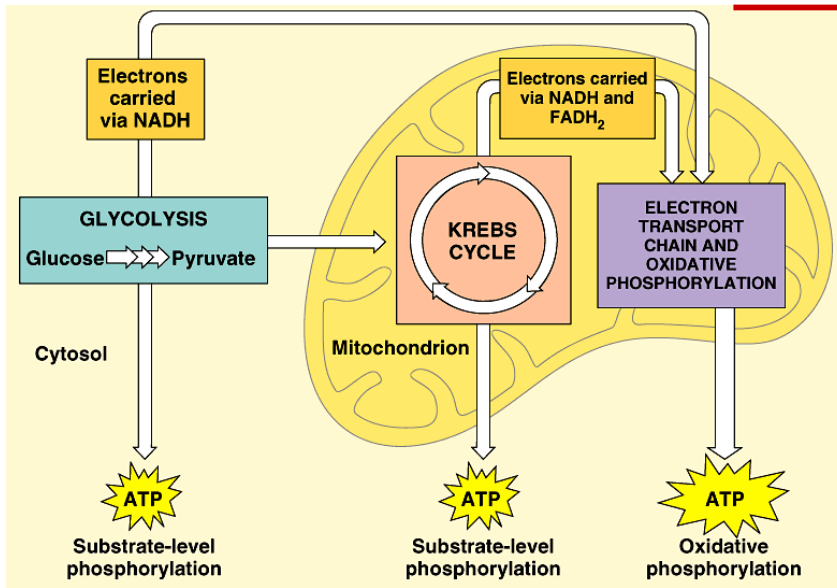
Energy accounting of Krebs cycle

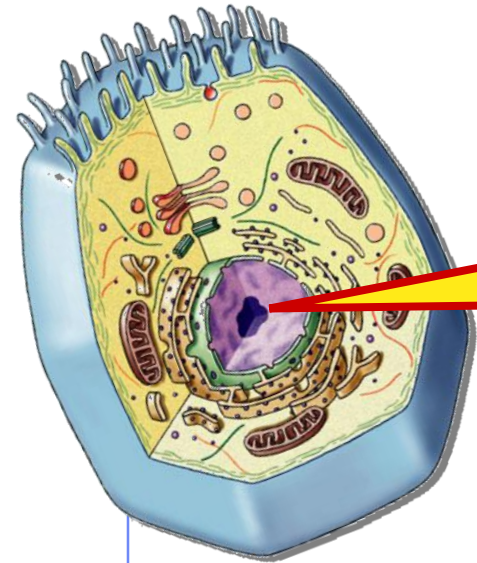


Net gain = 2 ATP
= 8 NADH + 2 FADH₂

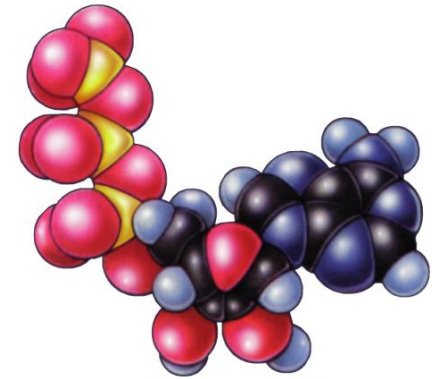
Value of Krebs cycle?

- If the yield is only 2 ATP then how was the Krebs cycle an adaptation?
 - ◆ value of NADH & FADH₂
 - electron carriers & H carriers
 - ◆ reduced molecules move electrons
 - ◆ reduced molecules move H⁺ ions
 - to be used in the Electron Transport Chain



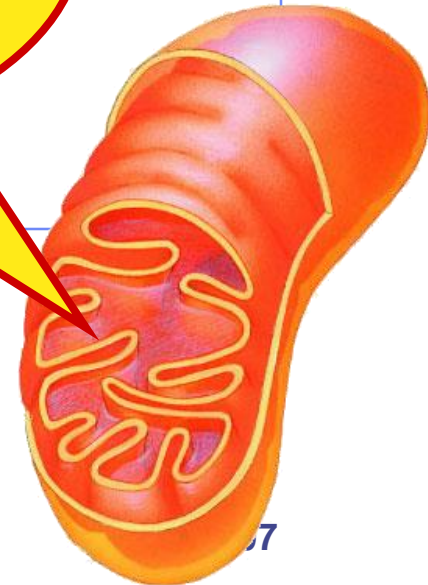


What's the point?



The point is to make **ATP!**

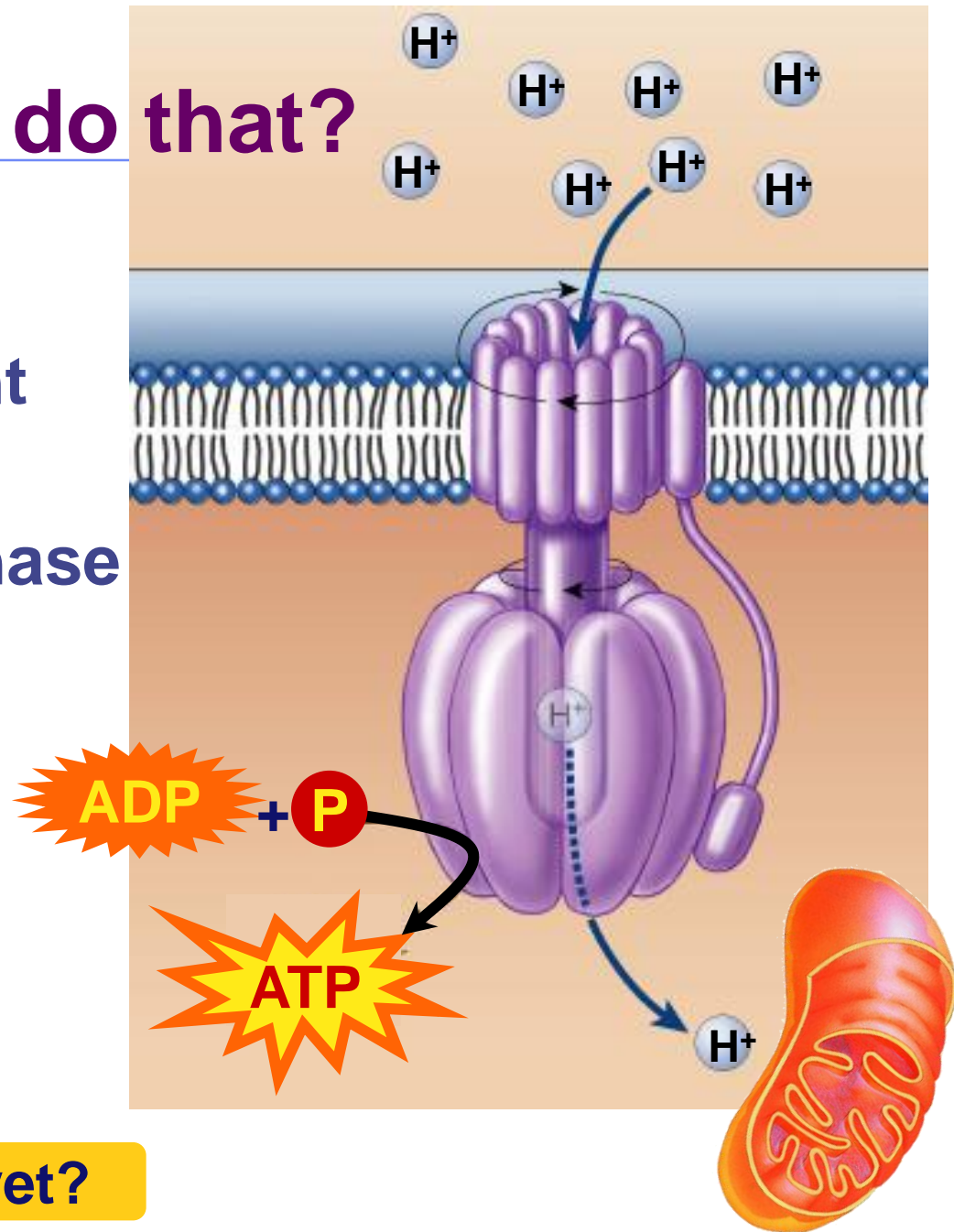
ATP



And how do we do that?

■ ATP synthase

- ◆ set up a H^+ gradient
- ◆ allow H^+ to flow through ATP synthase
- ◆ powers bonding of P_i to ADP



But... Have we done that yet?

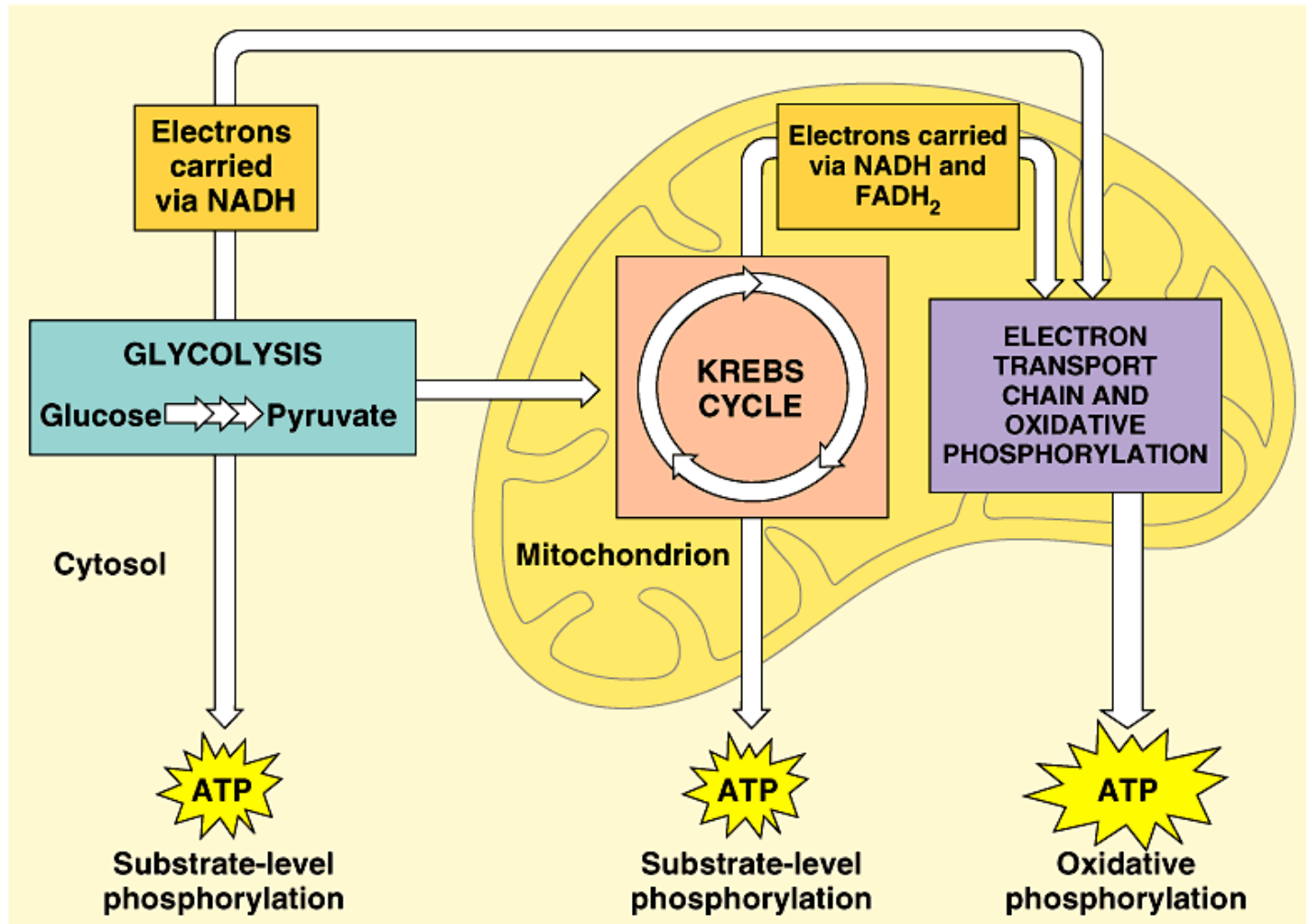


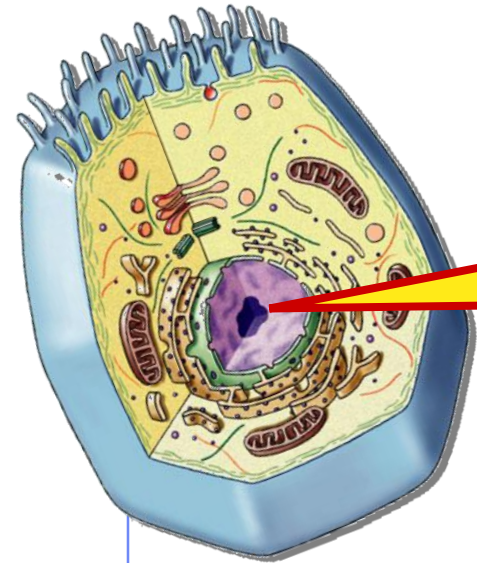
NO!
The final chapter
to my story is
next!
Any Questions?

Cellular Respiration
Stage 4:
Electron Transport Chain

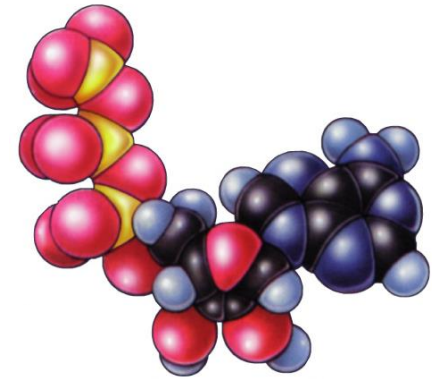


Cellular respiration



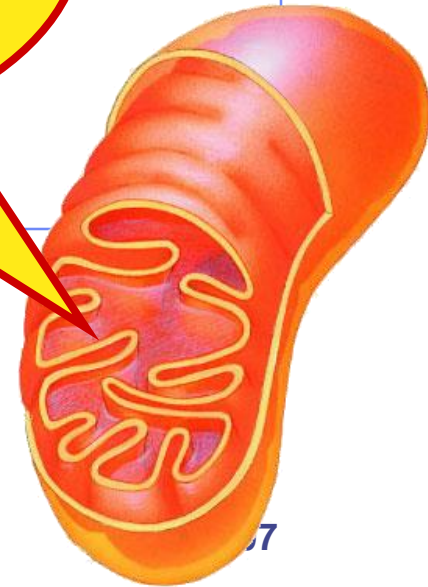


What's the point?



The point is to make **ATP!**

ATP

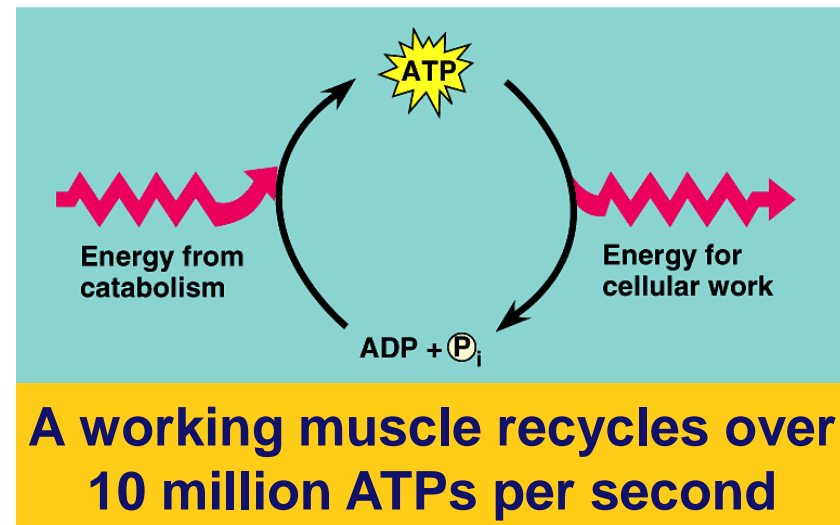


ATP accounting so far...

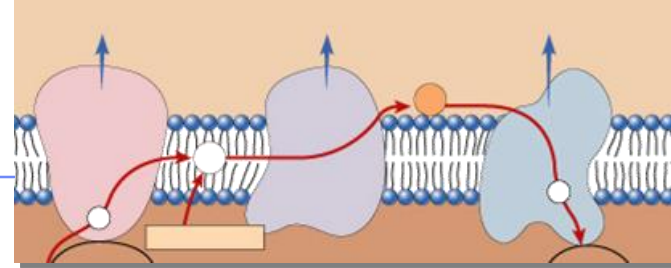
- Glycolysis → **2 ATP**
- Kreb's cycle → **2 ATP**
- Life takes a lot of energy to run, need to extract more energy than **4 ATP!**

There's got to be a better way!

I need a lot more ATP!

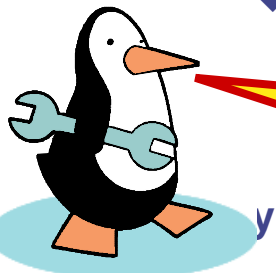


There *is* a better way!



▪ Electron Transport Chain

- ◆ series of proteins built into inner mitochondrial membrane
 - along cristae
 - transport proteins & enzymes
- ◆ transport of electrons down ETC linked to pumping of H^+ to create H^+ gradient
- ◆ yields ~36 ATP from 1 glucose!
- ◆ only in presence of O_2 (aerobic respiration)

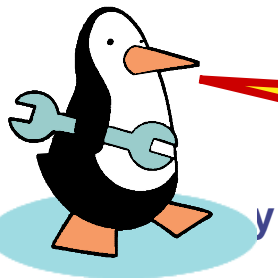
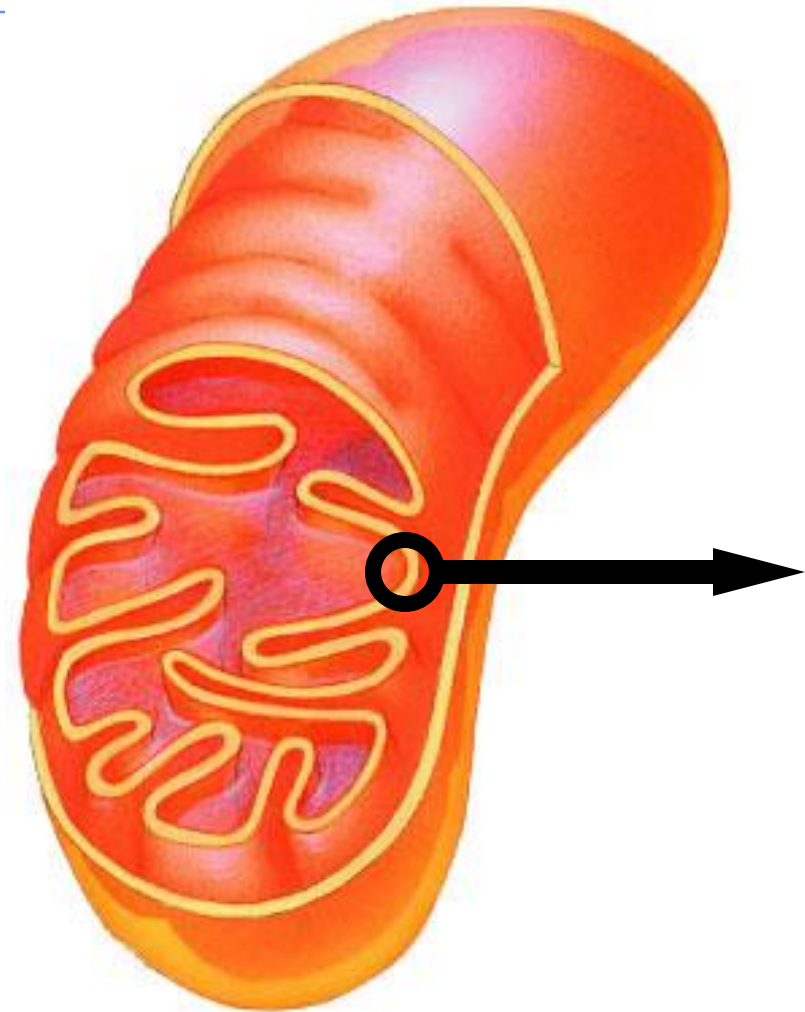


That sounds more like it!



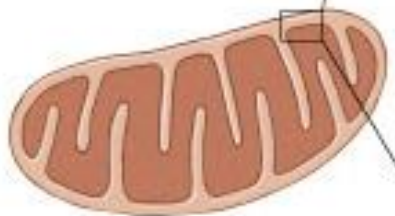
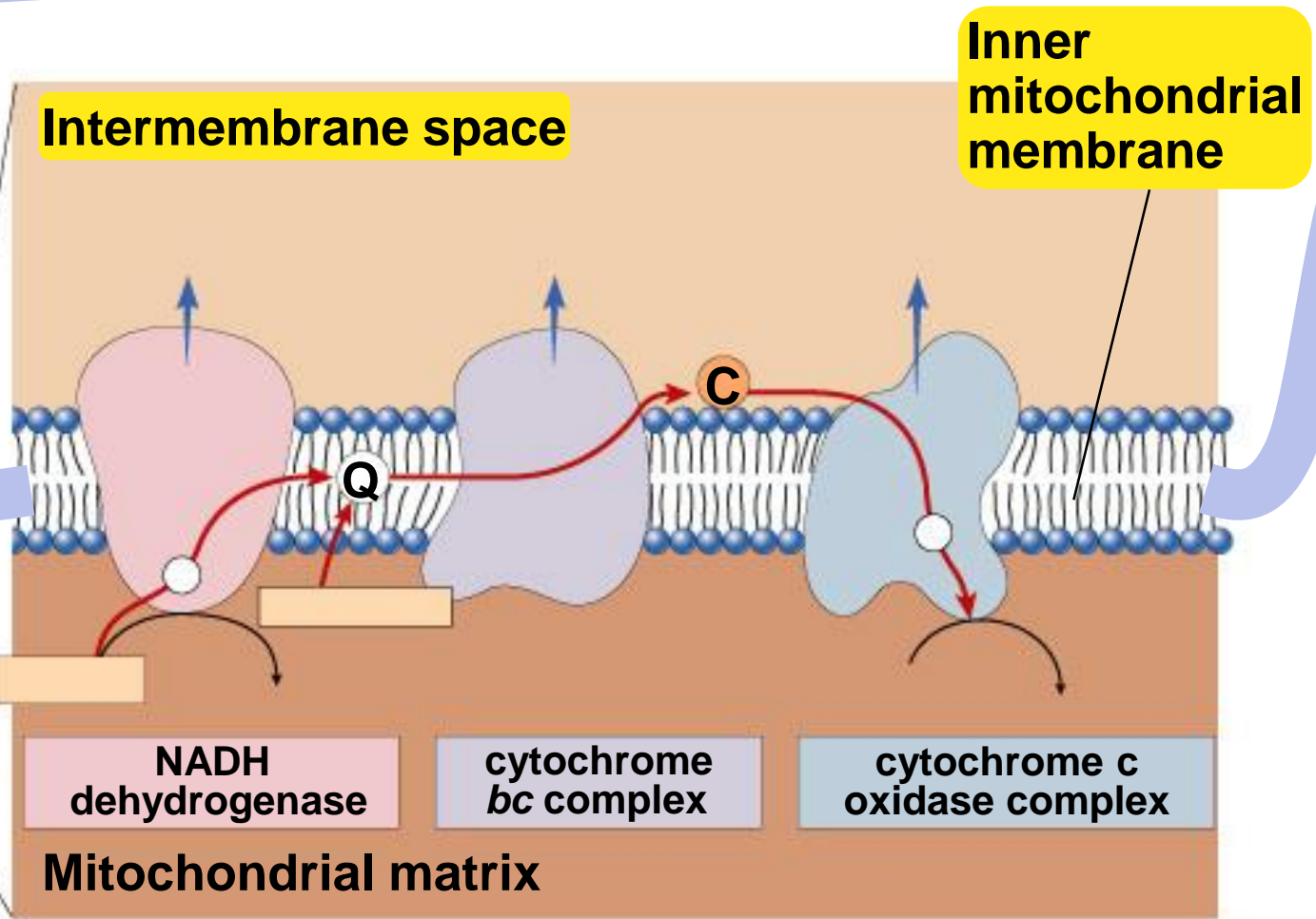
Mitochondria

- **Double membrane**
 - ◆ outer membrane
 - ◆ **inner membrane**
 - highly folded **cristae**
 - enzymes & transport proteins
 - ◆ **intermembrane space**
 - fluid-filled space between membranes



Ooooooh!
Form fits
function!

Electron Transport Chain

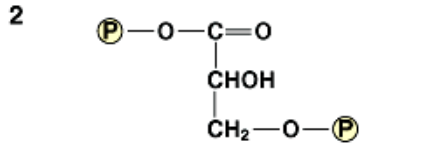
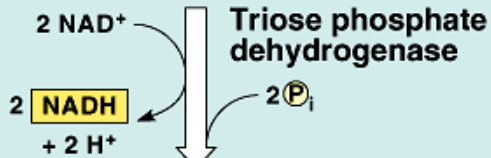


Remember the Electron Carriers?

Glycolysis

glucose

G3P

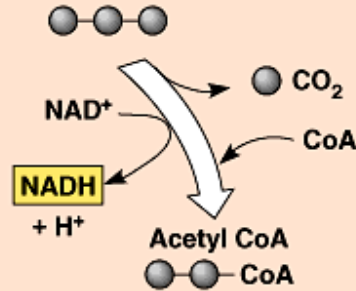


1, 3-Bisphosphoglycerate

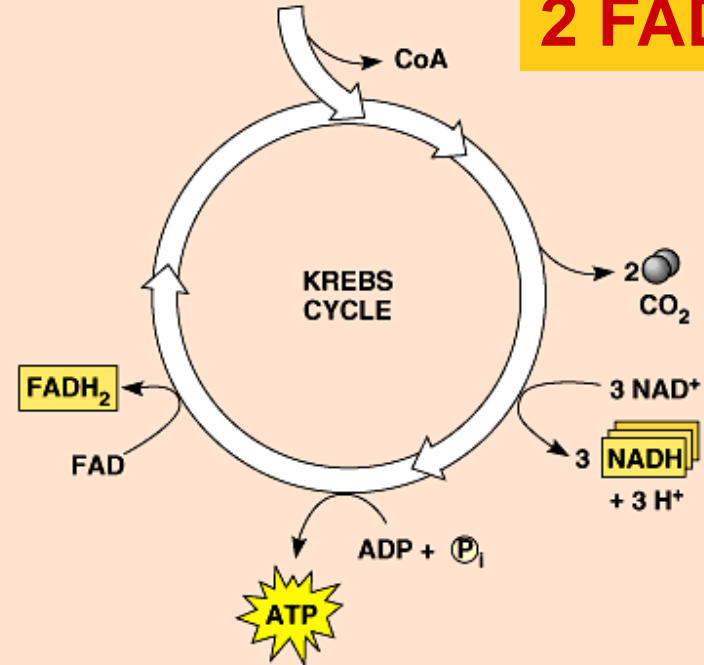
2 NADH

Krebs cycle

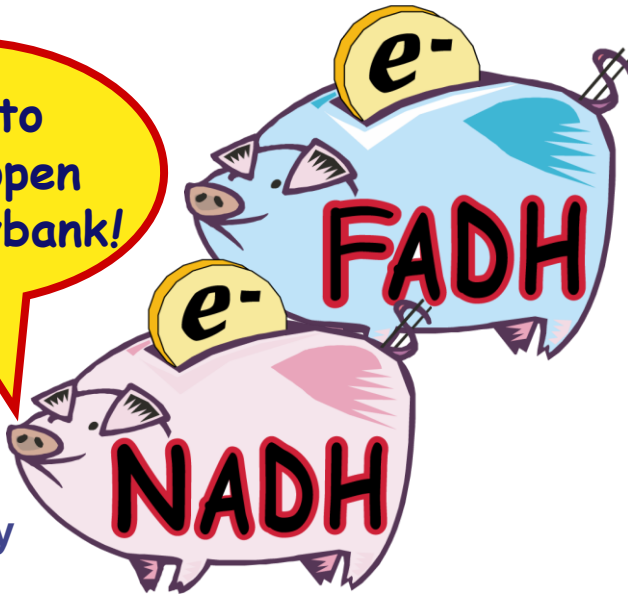
Pyruvate (from glycolysis, 2 molecules per glucose)



8 NADH
2 FADH₂



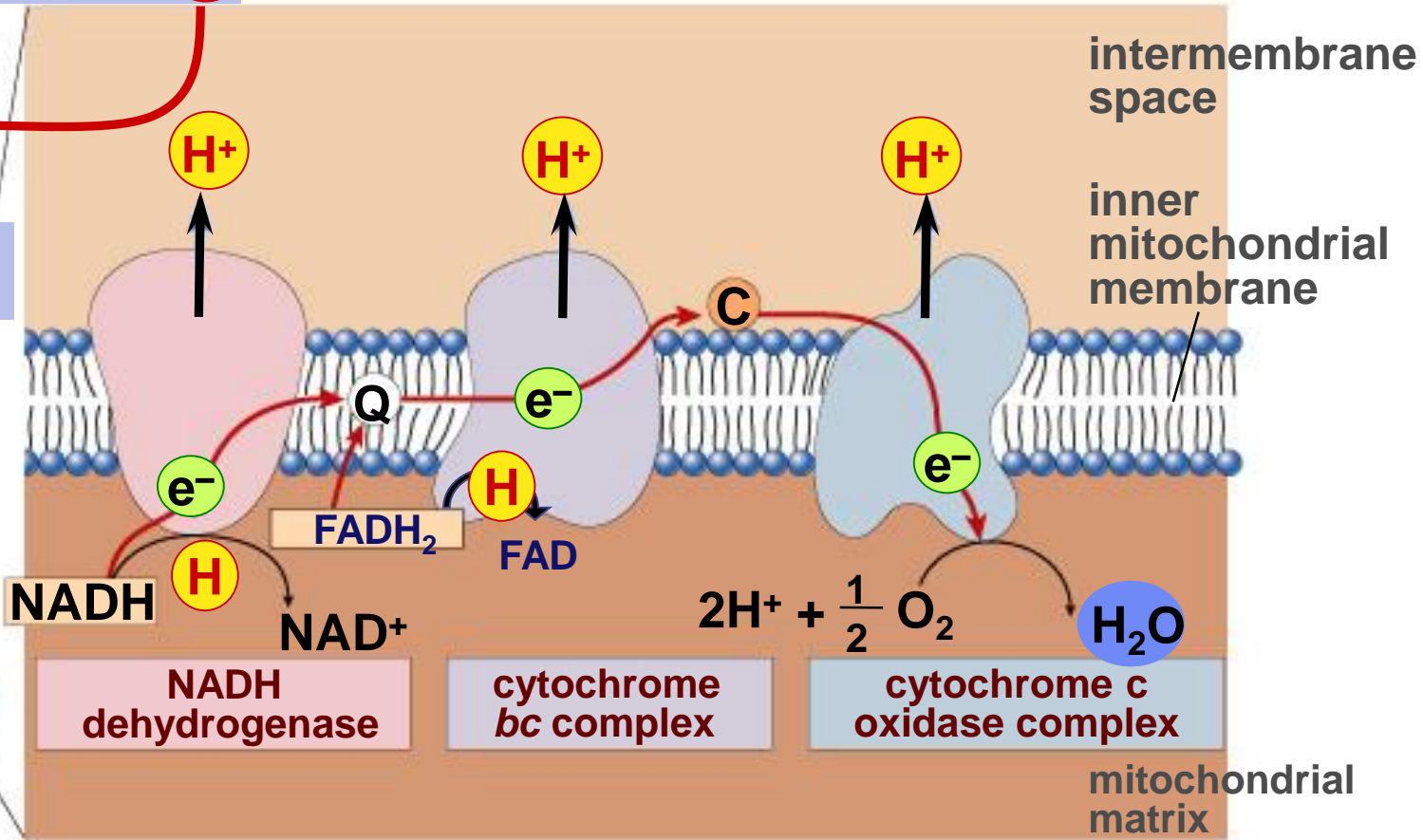
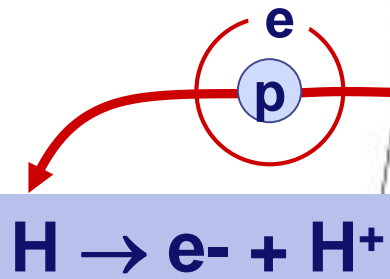
Time to break open the piggybank!



Electron Transport Chain



Building proton gradient!

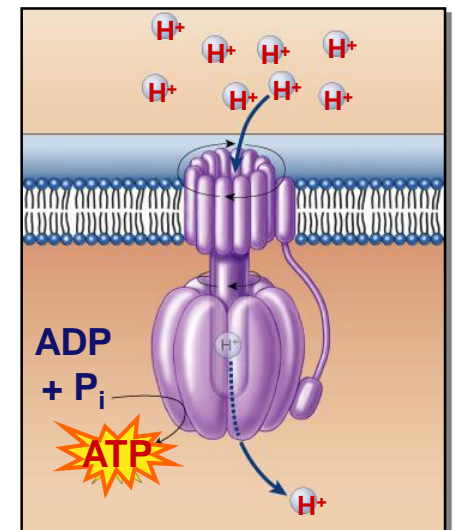
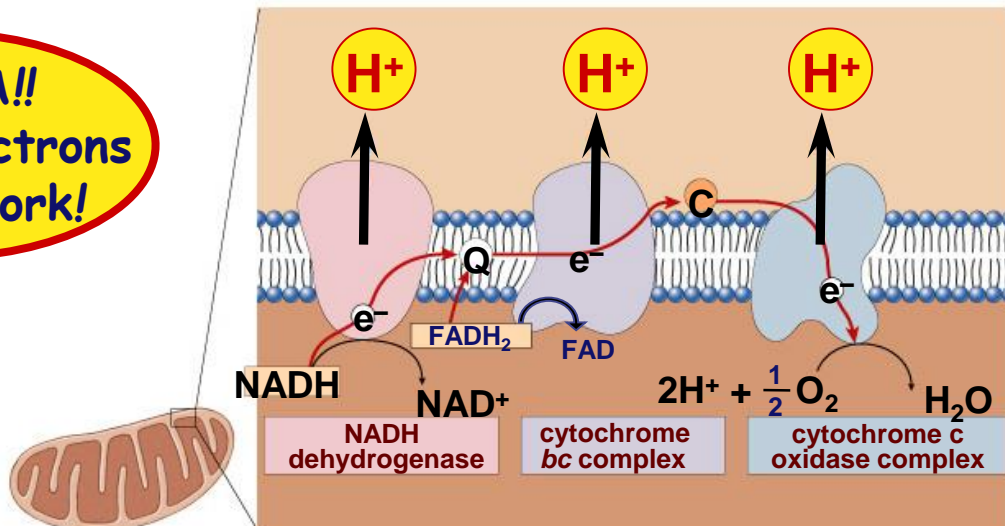
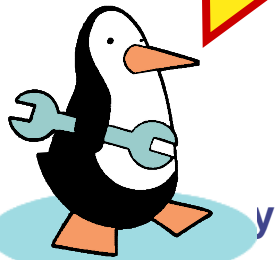


What powers the proton (H^+) pumps?...

Stripping H from Electron Carriers

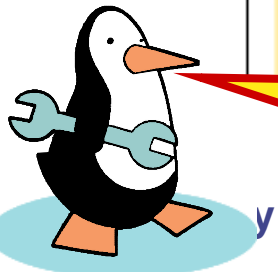
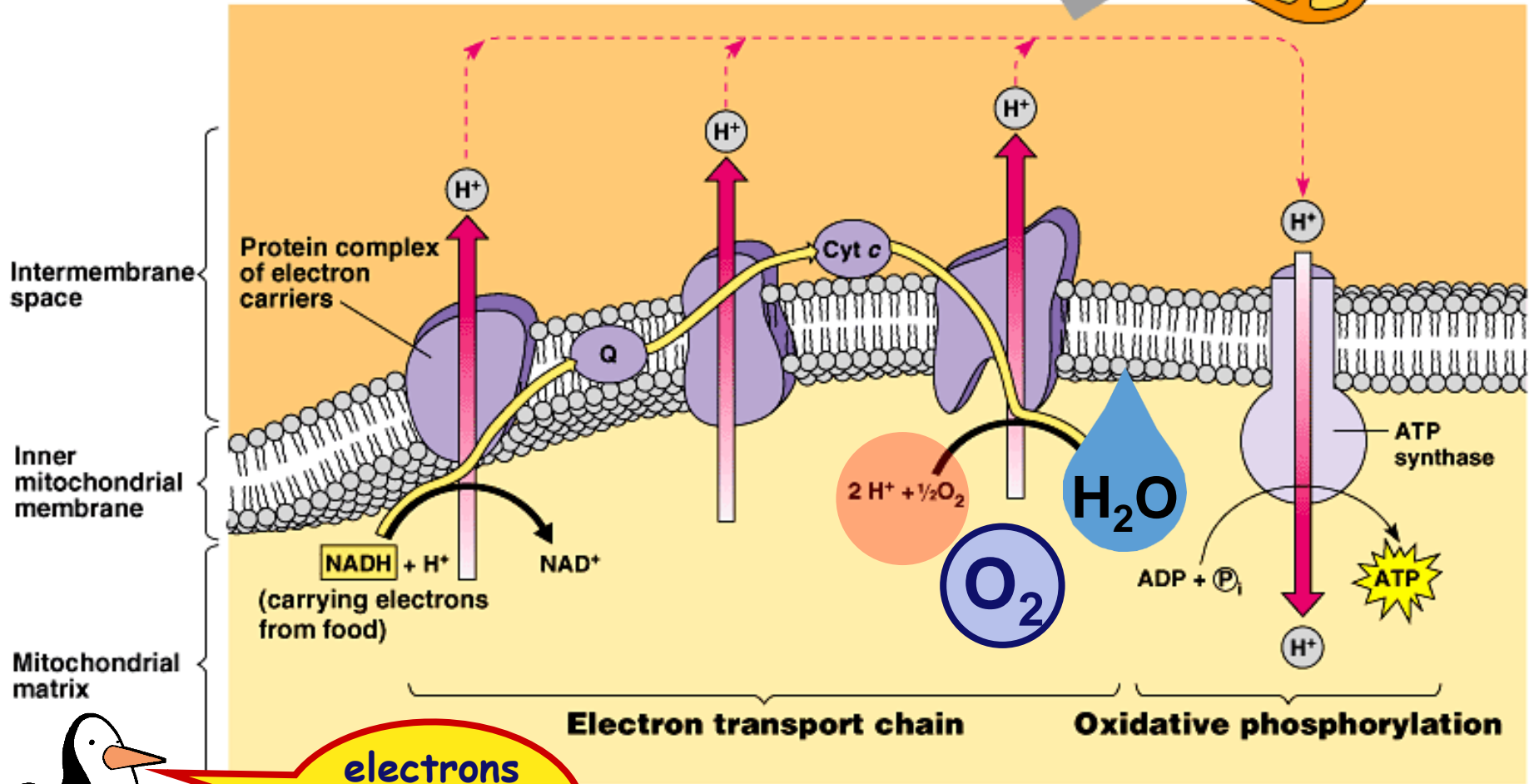
- **Electron carriers pass electrons & H⁺ to ETC**
 - ◆ H cleaved off NADH & FADH₂
 - ◆ **electrons** stripped from H atoms → **H⁺ (protons)**
 - electrons passed from one electron carrier to next in mitochondrial membrane (ETC)
 - flowing electrons = energy to do work
 - ◆ transport proteins in membrane pump **H⁺ (protons)** across inner membrane to **intermembrane space**

TA-DA!!
Moving electrons
do the work!



But what "pulls" the electrons down the ETC?

Inner mitochondrial membrane



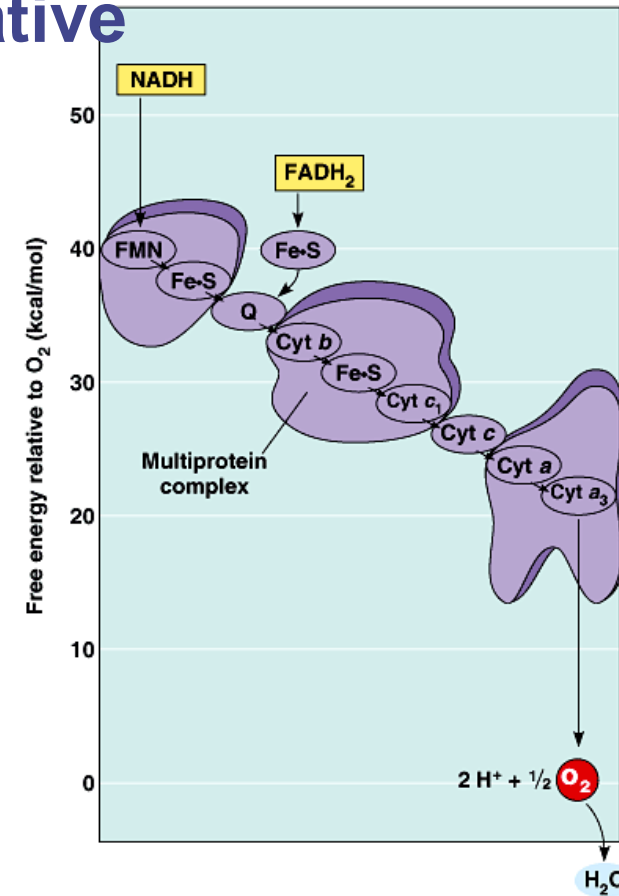
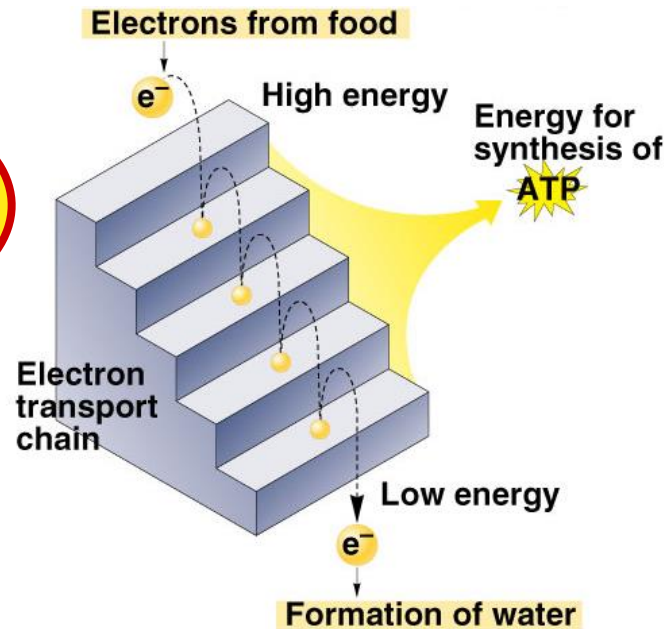
electrons flow downhill to O₂

oxidative phosphorylation

Electrons flow downhill

- Electrons move in steps from carrier to carrier downhill to **oxygen**
 - ◆ each carrier more electronegative
 - ◆ controlled oxidation
 - ◆ controlled release of energy

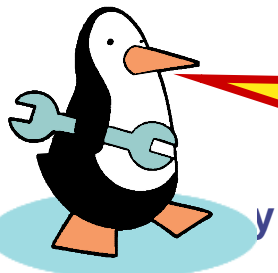
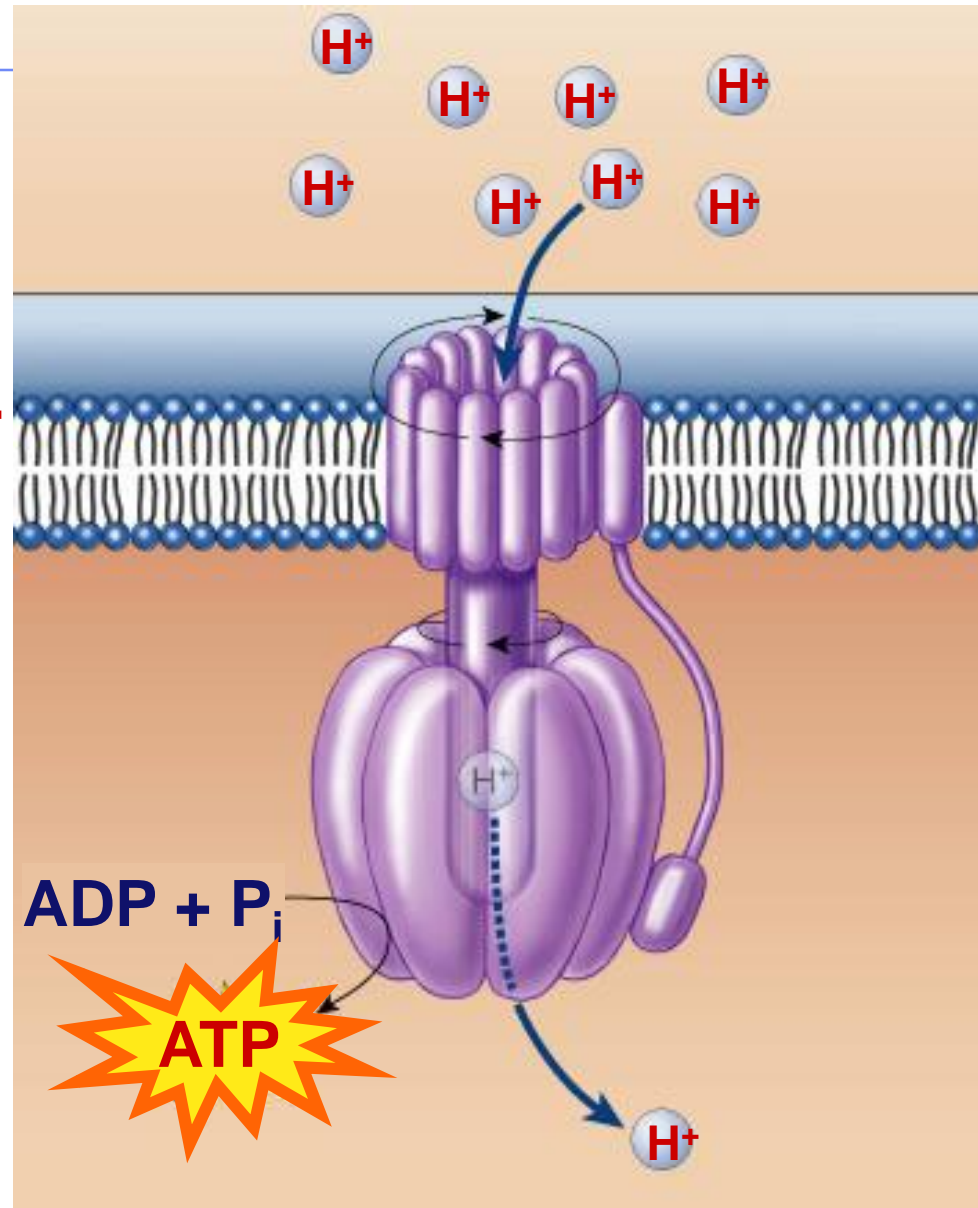
make ATP instead of fire!



“proton-motive” force

We did it!

- Set up a H^+ gradient
- Allow the protons to flow through ATP synthase
- Synthesizes ATP

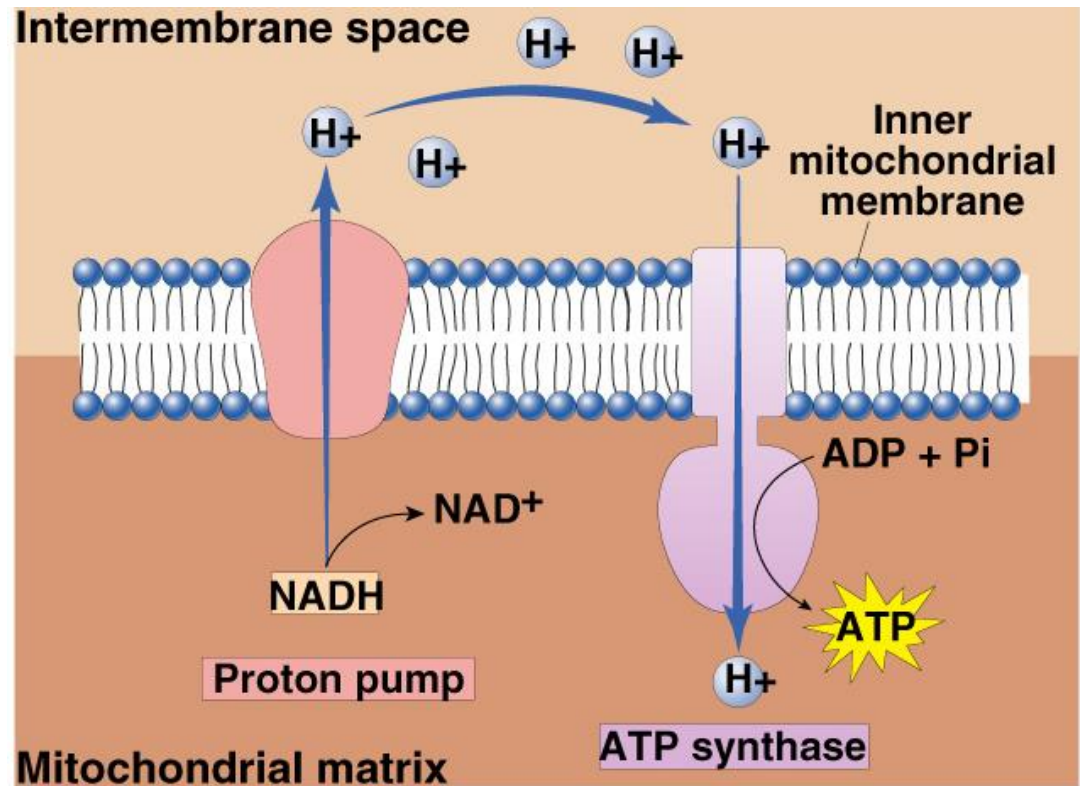


Are we there yet?

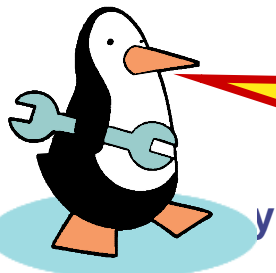
Chemiosmosis

- The diffusion of ions across a membrane
 - ◆ build up of proton gradient just so H^+ could flow through ATP synthase enzyme to build ATP

Chemiosmosis
links the Electron
Transport Chain
to ATP synthesis



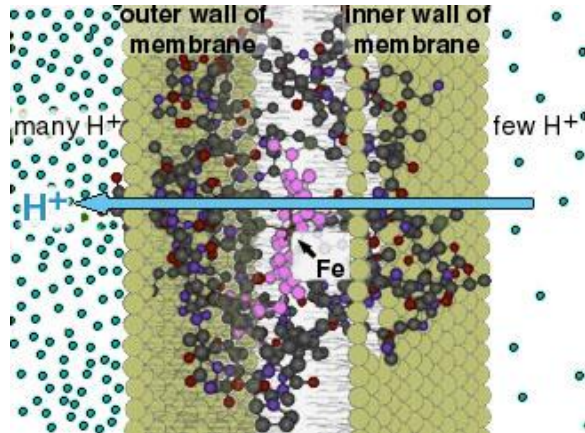
So that's
the point!



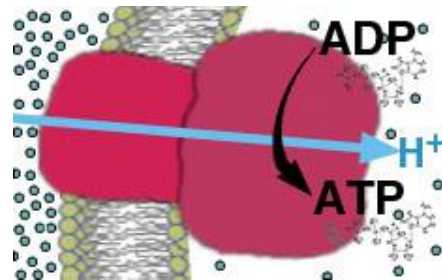
1961 | 1978

Peter Mitchell

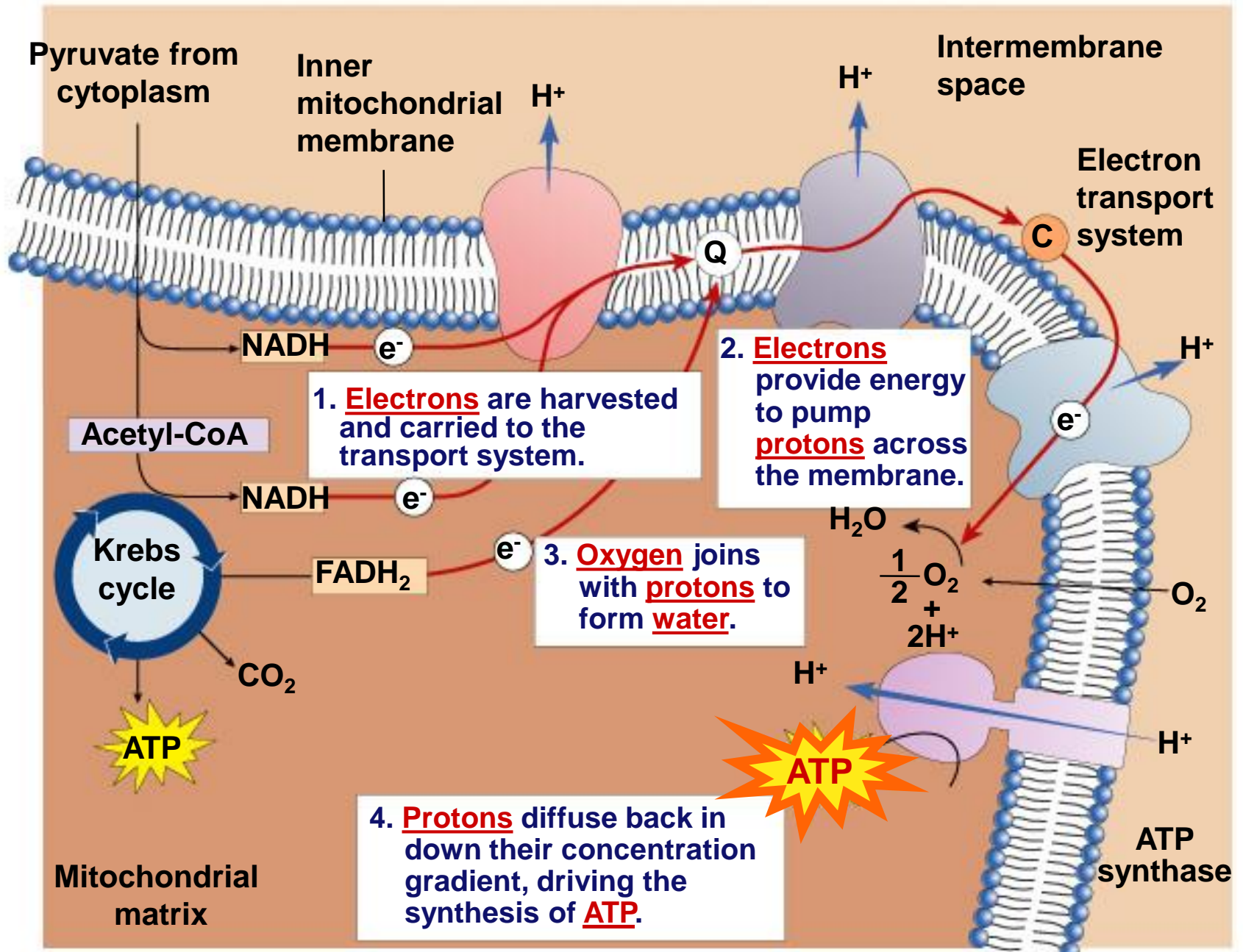
- Proposed chemiosmotic hypothesis
 - ◆ revolutionary idea at the time



proton motive force



1920-1992

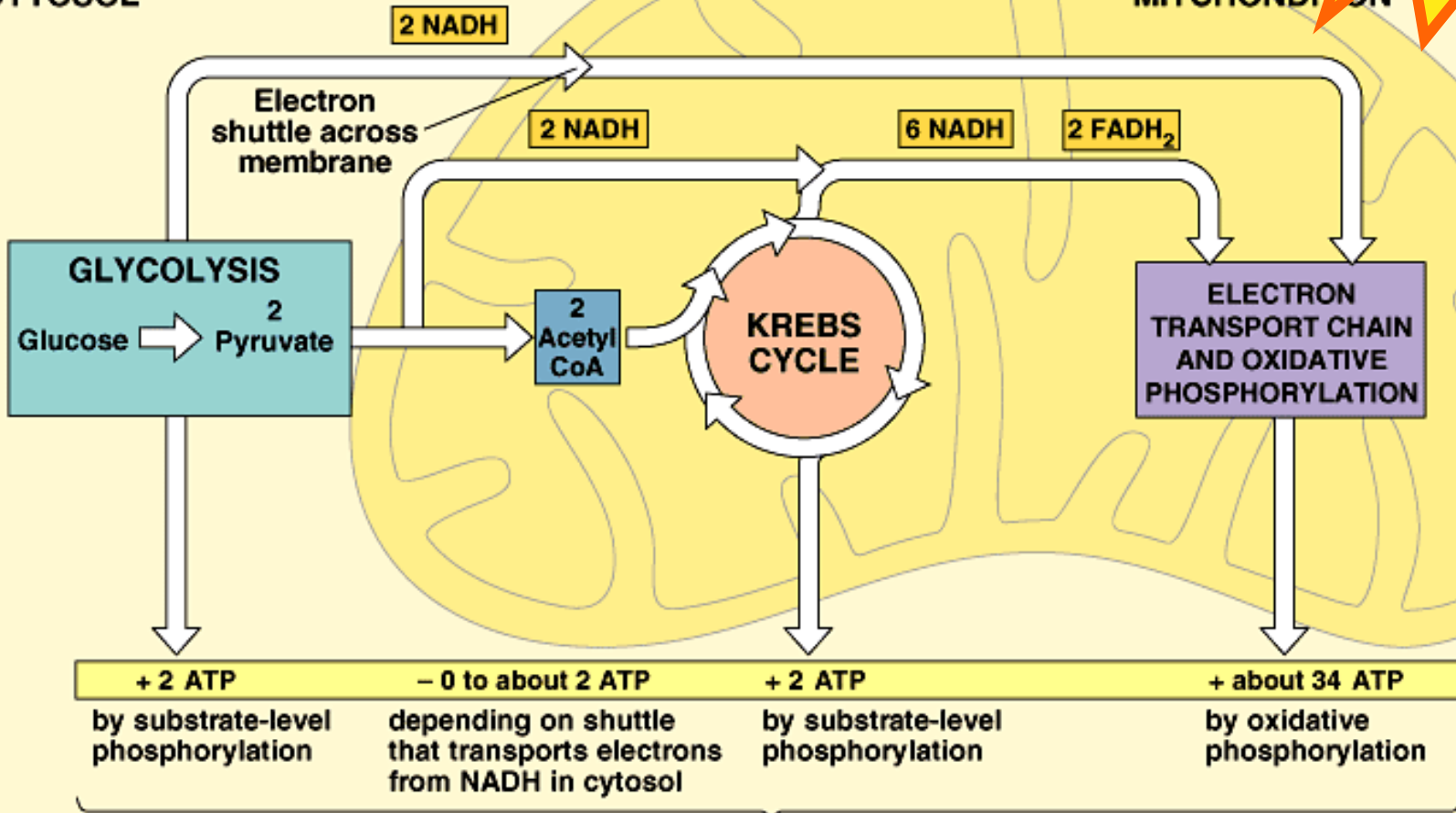


Cellular respiration

~40 ATP

CYTOSOL

MITCHONDRION



2 ATP

+

2 ATP

+

~36 ATP

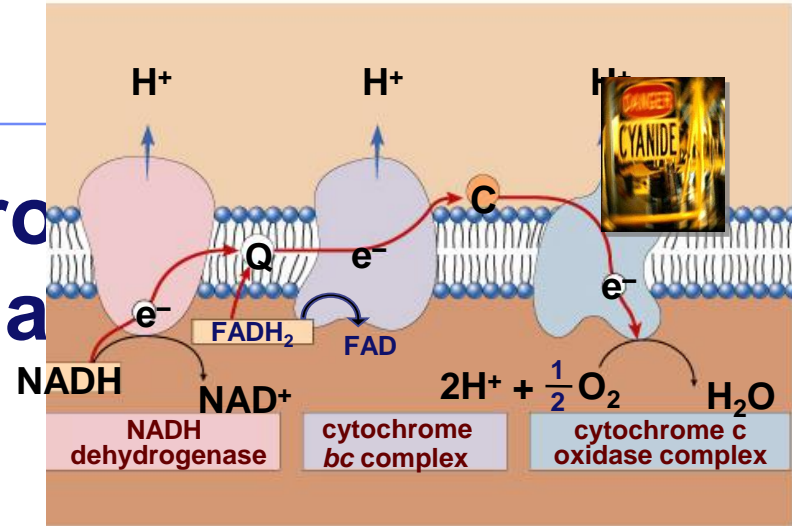
Summary of cellular respiration



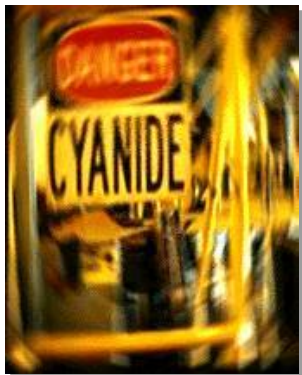
- Where did the glucose come from?
- Where did the O_2 come from?
- Where did the CO_2 come from?
- Where did the CO_2 go?
- Where did the H_2O come from?
- Where did the ATP come from?
- What else is produced that is not listed in this equation?
- Why do we breathe?

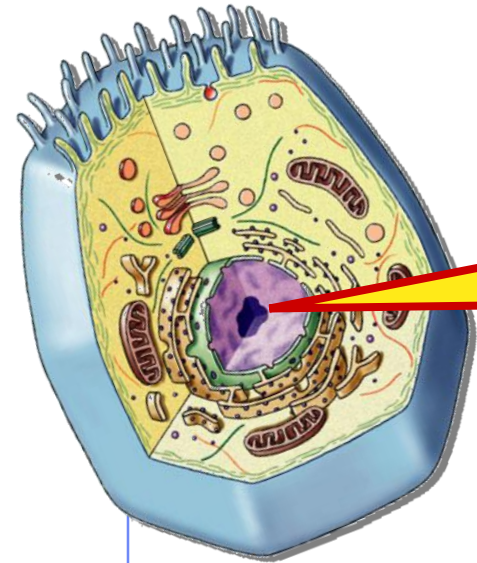
Taking it beyond...

- What is the final electron acceptor in the Electron Transport Chain?

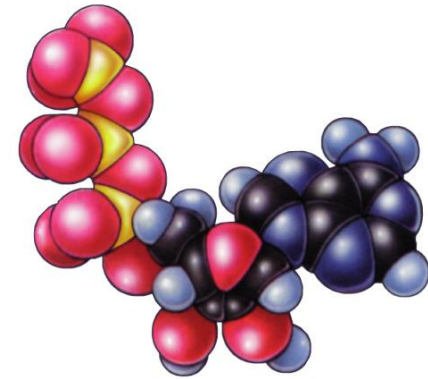


- So what happens if O_2 unavailable?
 - ETC backs up**
 - nothing to pull electrons down chain
 - NADH & $FADH_2$ can't unload H^+
 - ATP production ceases
 - cells run out of energy
 - and you die!



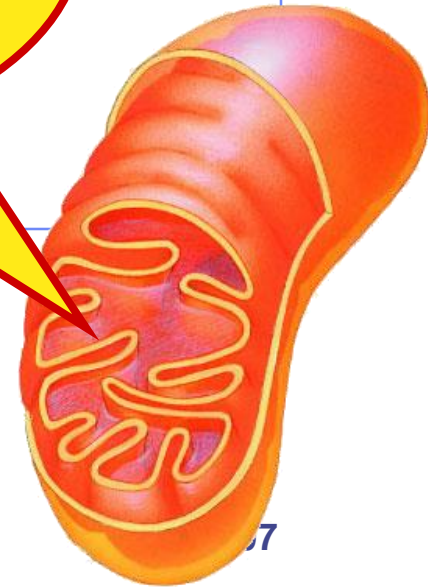


What's the point?



The point is to make **ATP!**

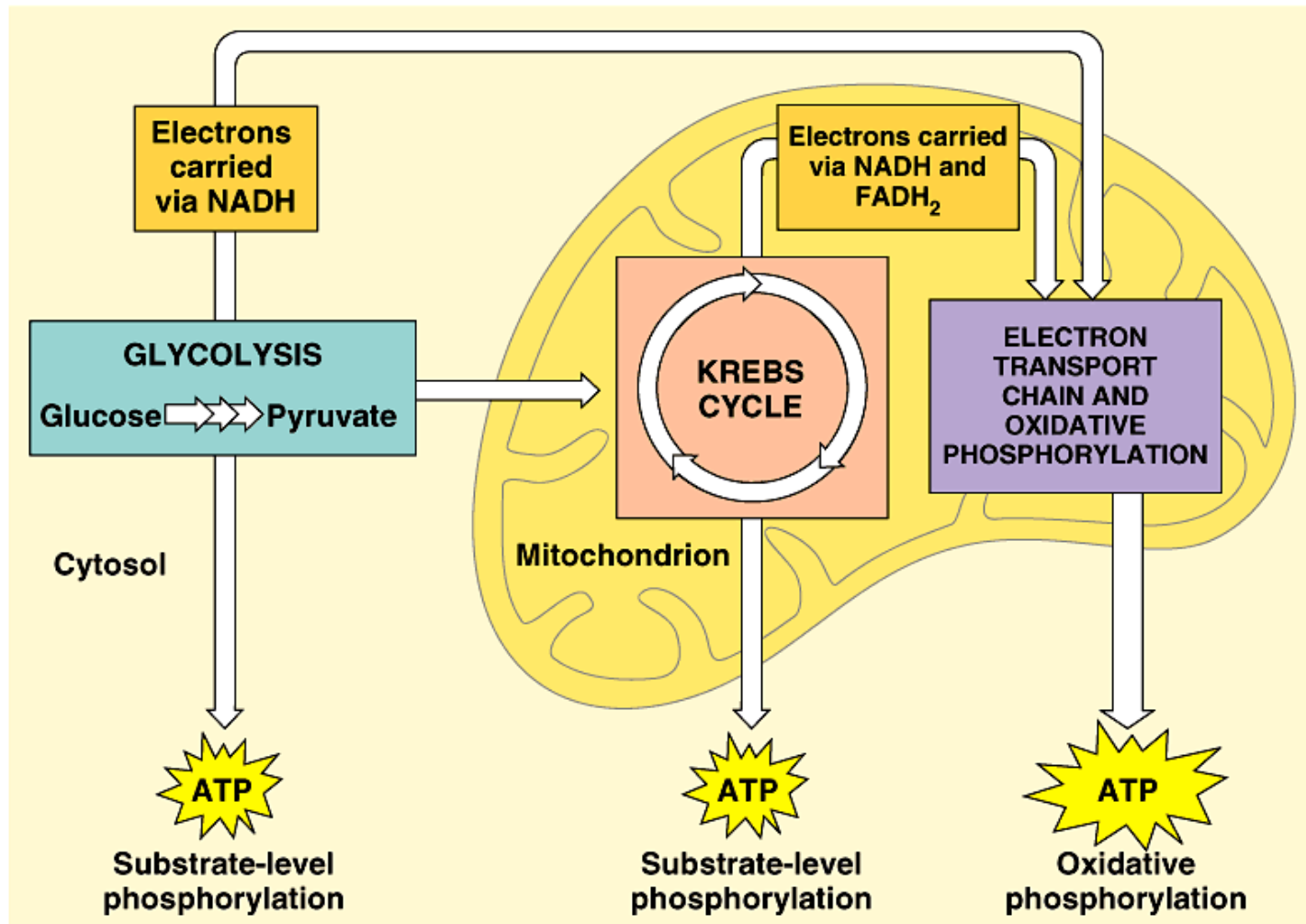
ATP



Chapter 9.
Cellular Respiration
Other Metabolites &
Control of Respiration



Cellular respiration



Beyond glucose: Other carbohydrates

- Glycolysis accepts a wide range of carbohydrates fuels

- ◆ polysaccharides → → → glucose
hydrolysis

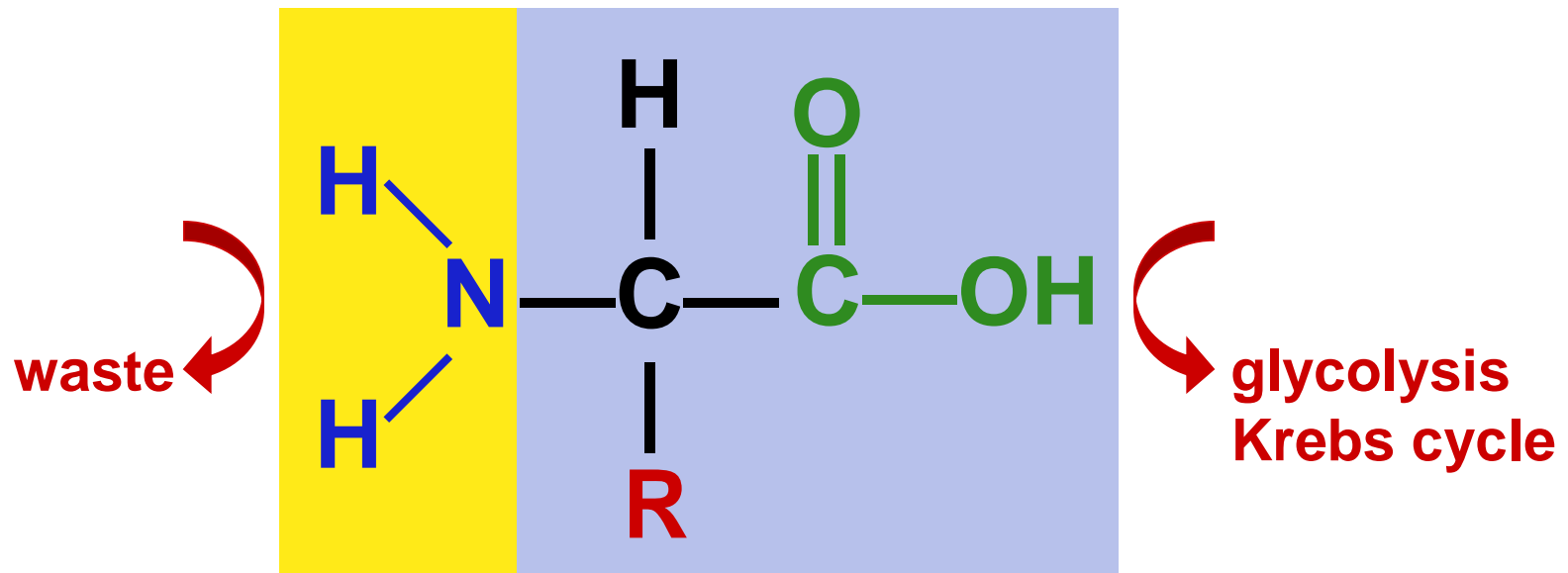
- ex. starch, glycogen

- ◆ other 6C sugars → → → glucose
modified

- ex. galactose, fructose

Beyond glucose: Proteins

- Proteins → → → → amino acids
hydrolysis

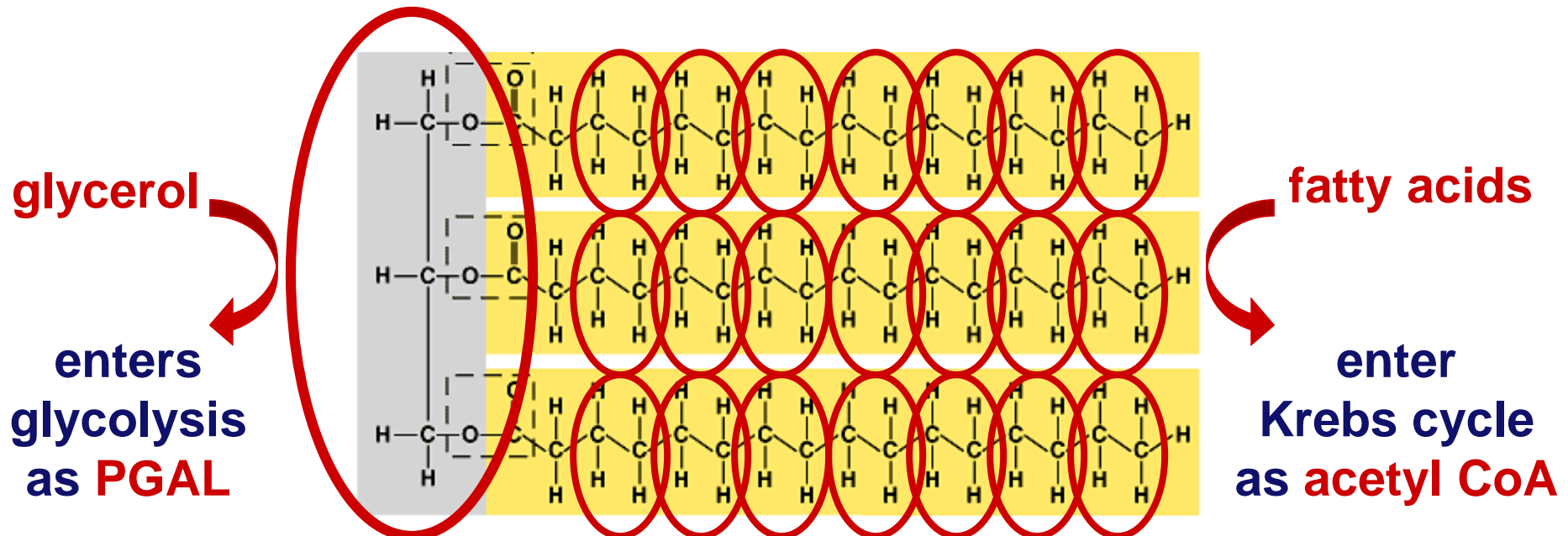


amino group =
waste product
excreted as
ammonia, urea,
or uric acid

carbon skeleton =
enters glycolysis
or Krebs cycle at
different stages

Beyond glucose: Fats

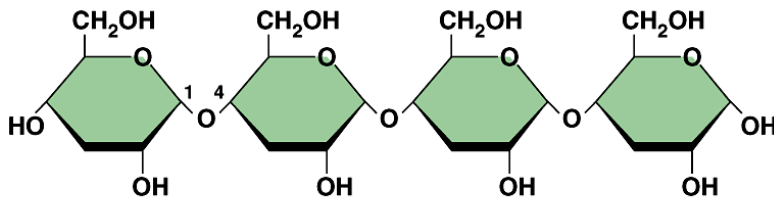
- Fats $\rightarrow \xrightarrow{\text{hydrolysis}} \rightarrow \rightarrow$ glycerol & fatty acids
 - ◆ glycerol (3C) $\rightarrow \rightarrow$ PGAL $\rightarrow \rightarrow$ glycolysis
 - ◆ fatty acids \rightarrow 2C acetyl groups \rightarrow acetyl coA \rightarrow Krebs cycle



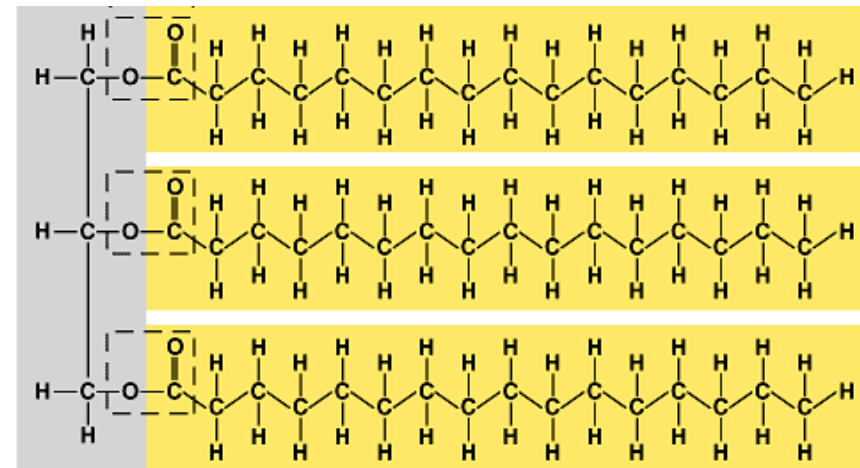
Carbohydrates vs. Fats

- Fat generates 2x **ATP** vs. carbohydrate
 - ◆ more **C** in gram of fat
 - ◆ more **O** in gram of carbohydrate
 - so it's already partly oxidized

carbohydrate

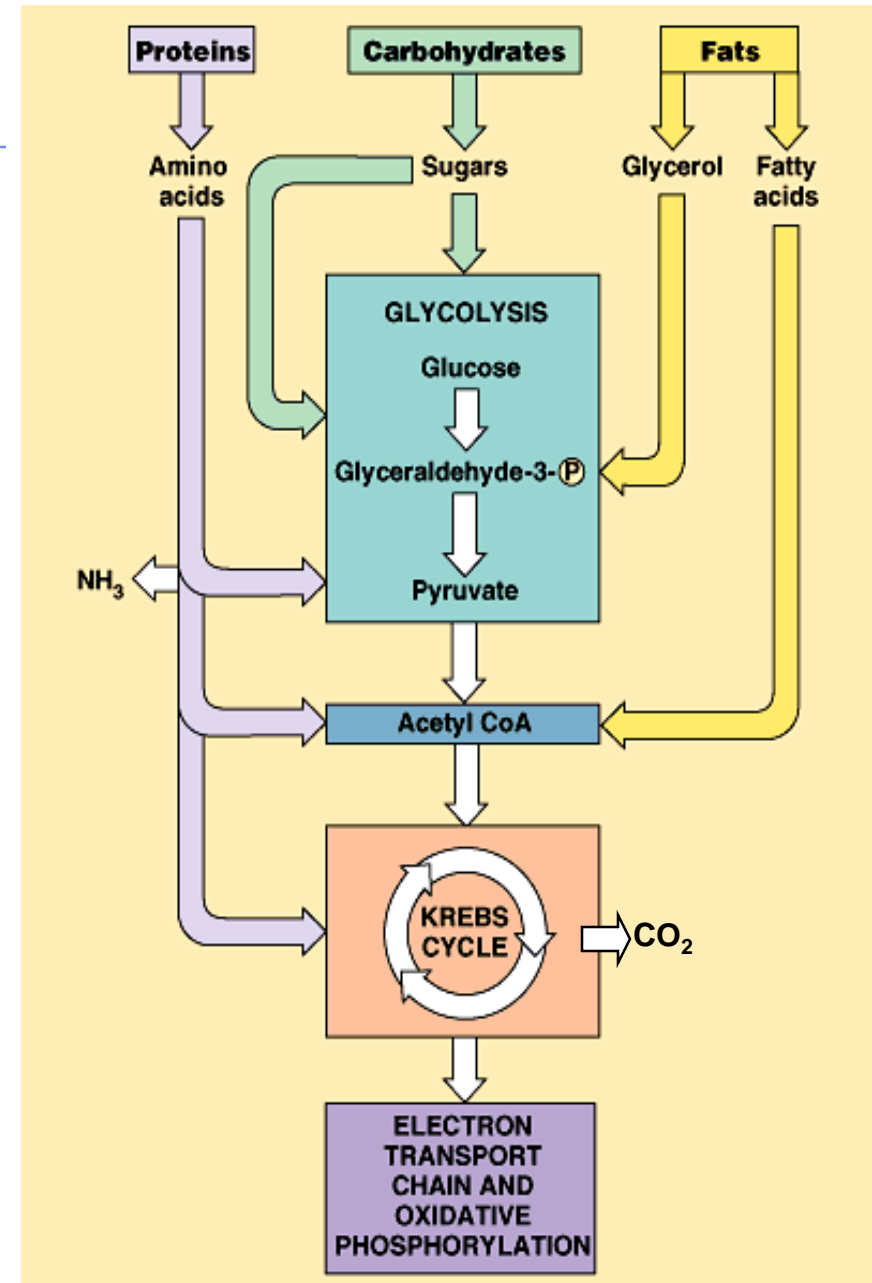


fat



Metabolism

- **Coordination of digestion & synthesis**
 - ◆ by regulating enzyme
- **Digestion**
 - ◆ digestion of carbohydrates, fats & proteins
 - all catabolized through same pathways
 - enter at different points
 - ◆ cell extracts energy from every source



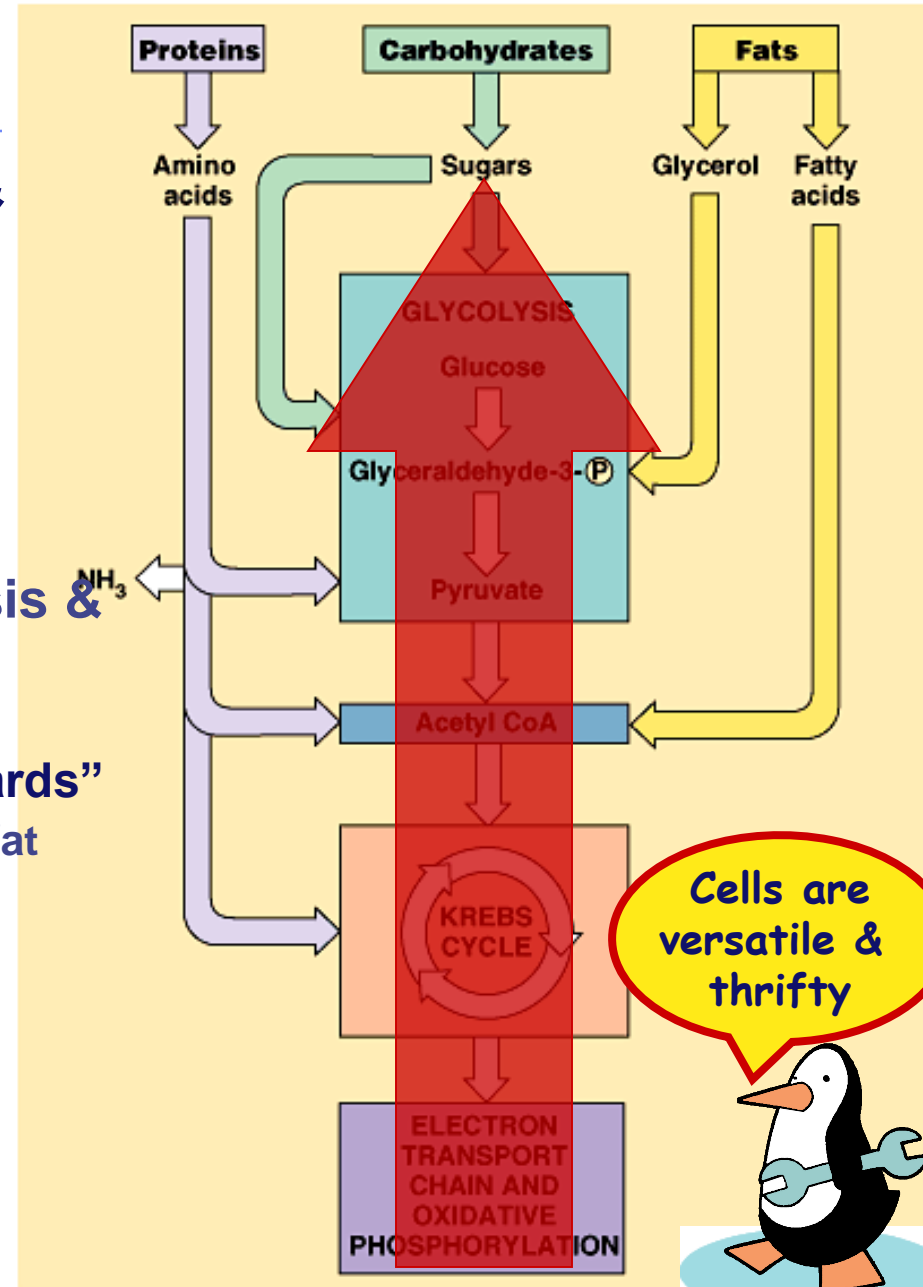
Metabolism

- **Coordination of digestion & synthesis**
 - ◆ by regulating enzyme
- **Synthesis**
 - ◆ enough energy? build stuff!
 - ◆ cell uses points in glycolysis & Krebs cycle as links to pathways for synthesis
 - run the pathways “backwards”
 - ◆ eat too much fuel, build fat

pyruvate → → glucose

Krebs cycle intermediaries → → amino acids

acetyl CoA → → fatty acids

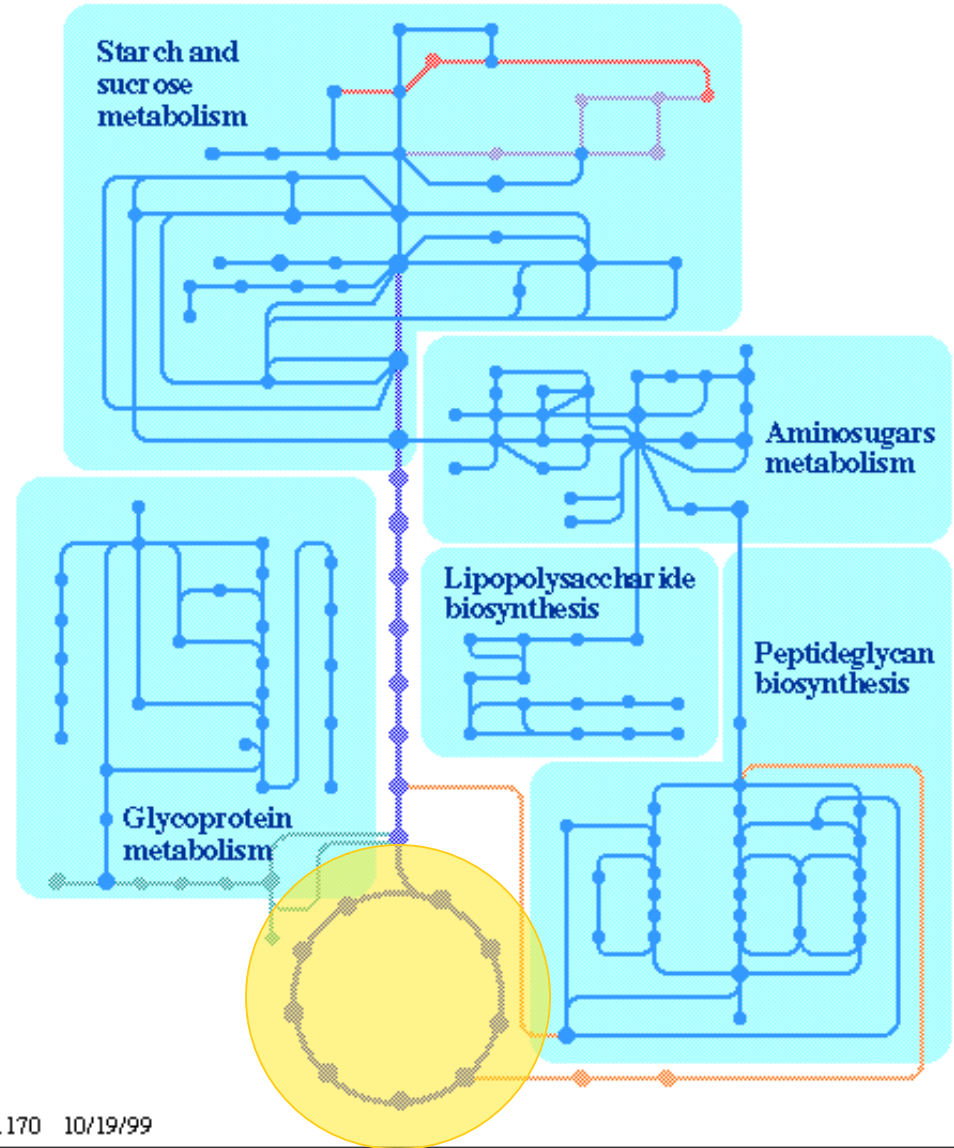


Carbohydrate Metabolism

- The many stops on the Carbohydrate Line

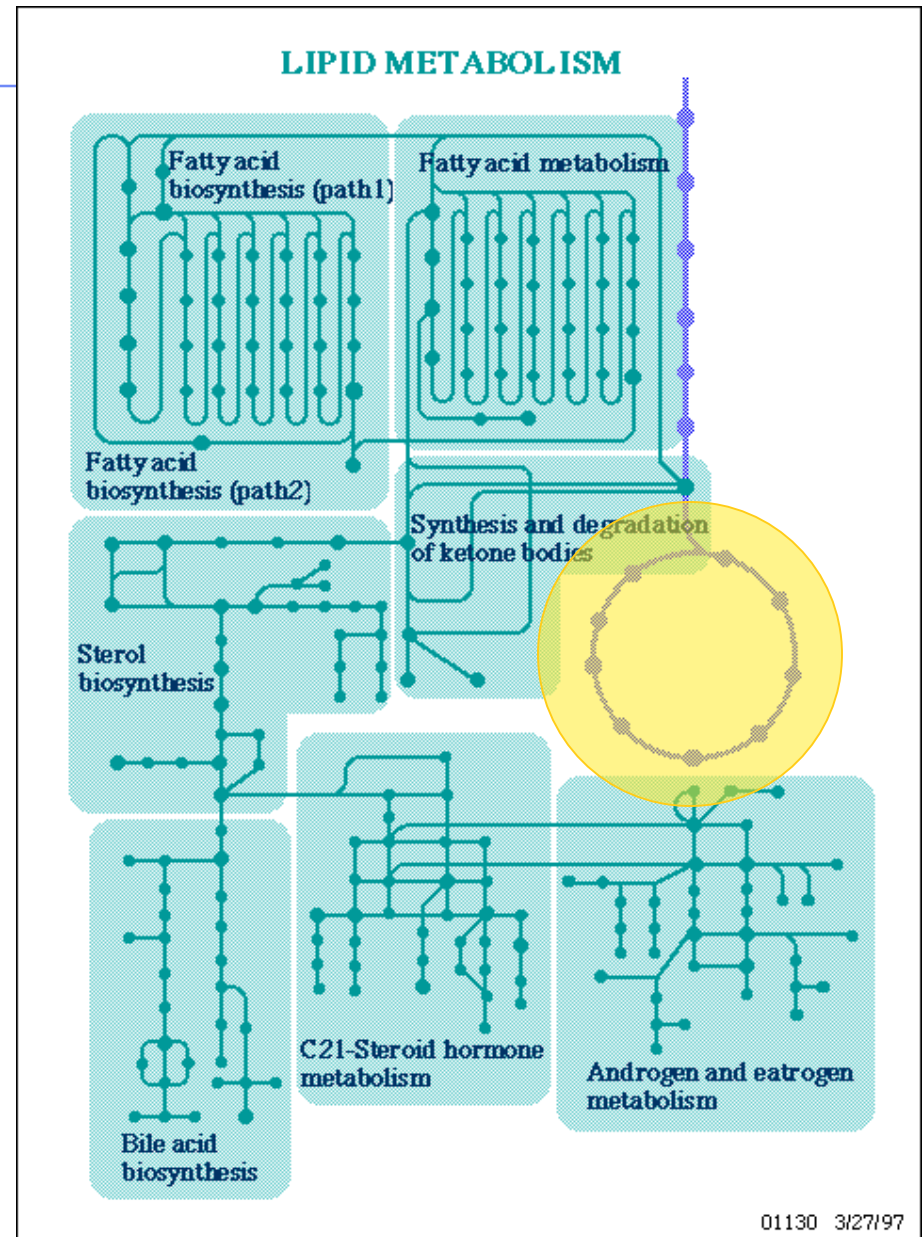
gluconeogenesis

METABOLISM OF COMPLEX CARBOHYDRATES



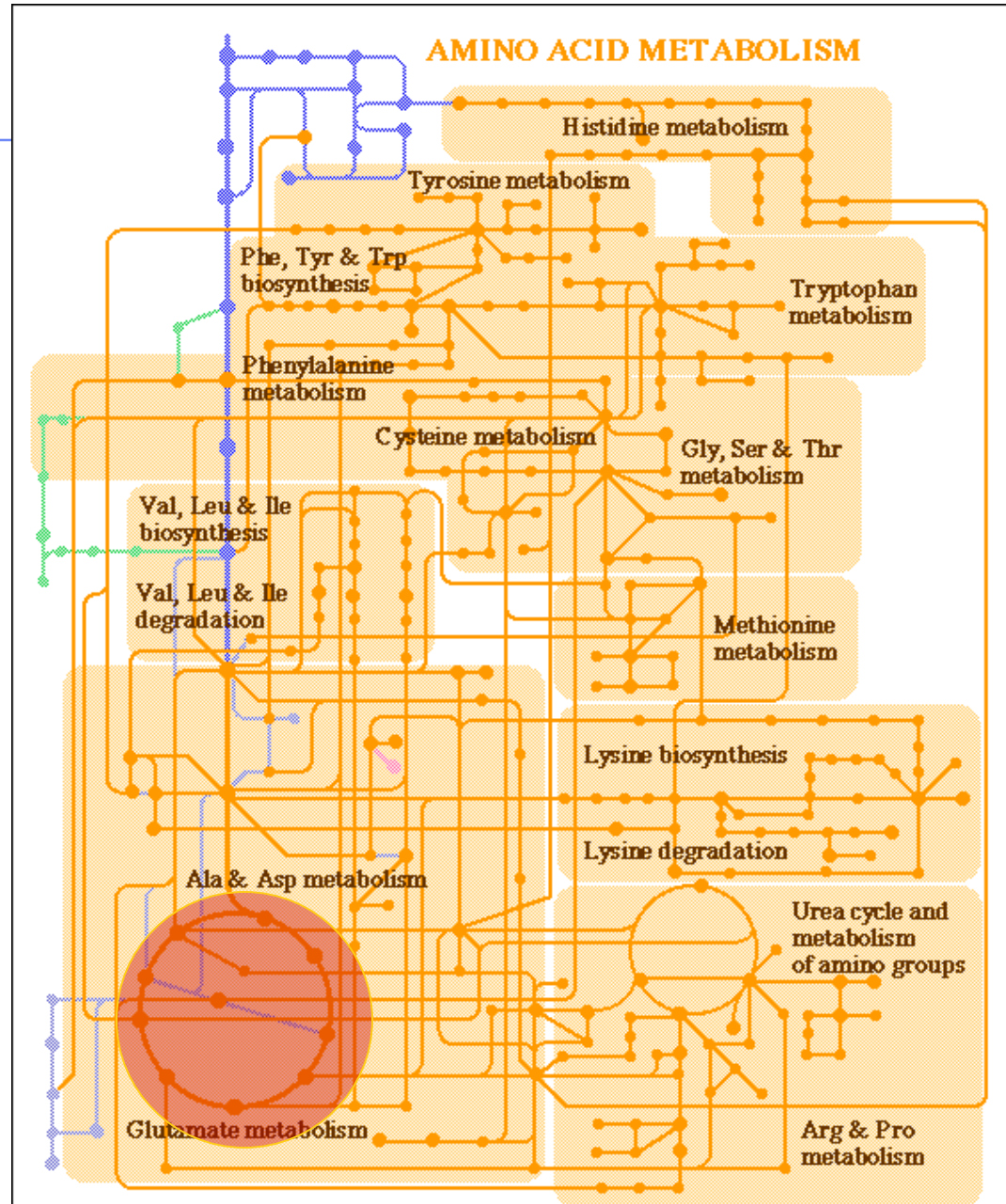
Lipid Metabolism

- The many stops on the Lipid Line



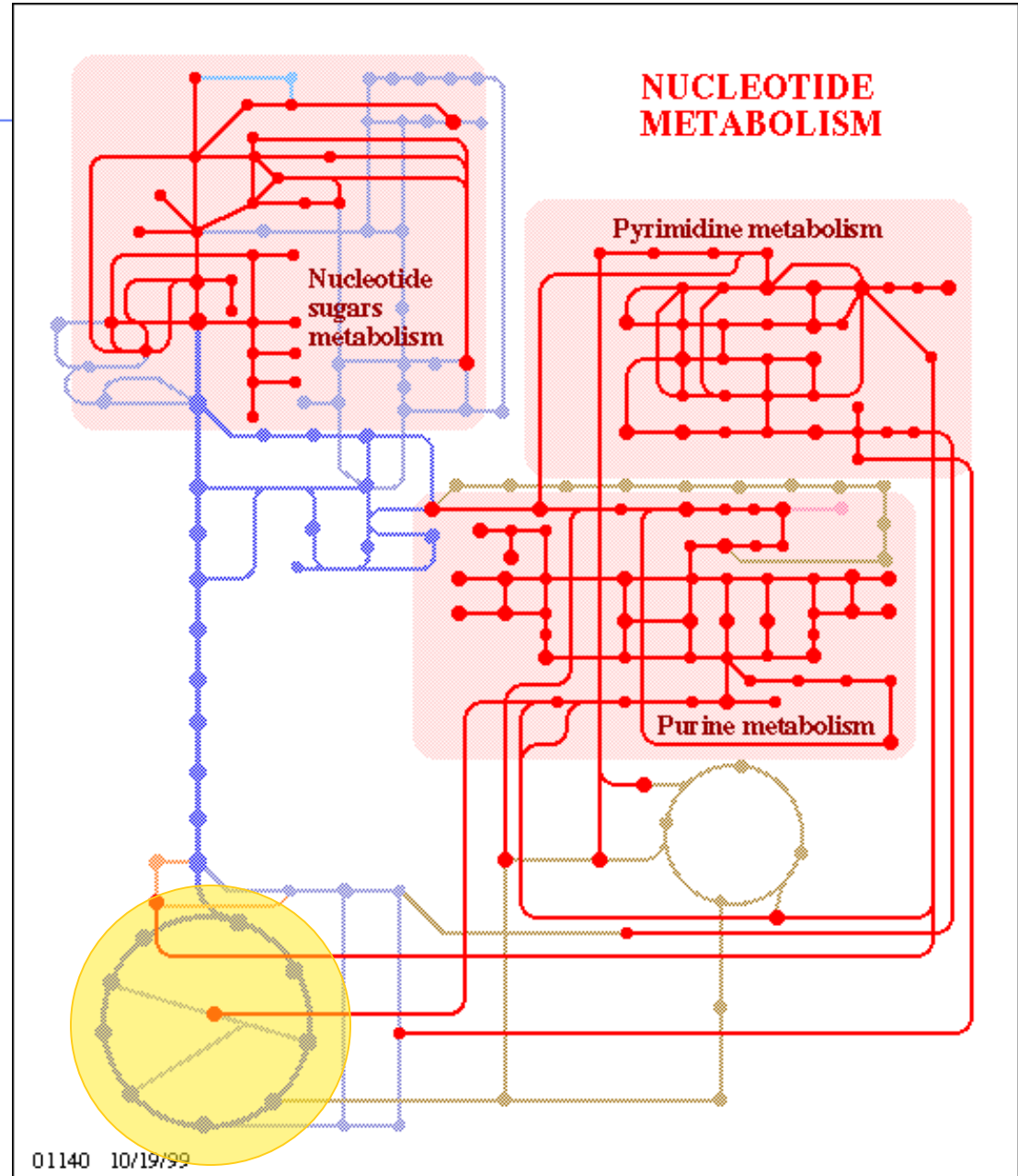
Amino Acid Metabolism

- The many stops on the AA Line



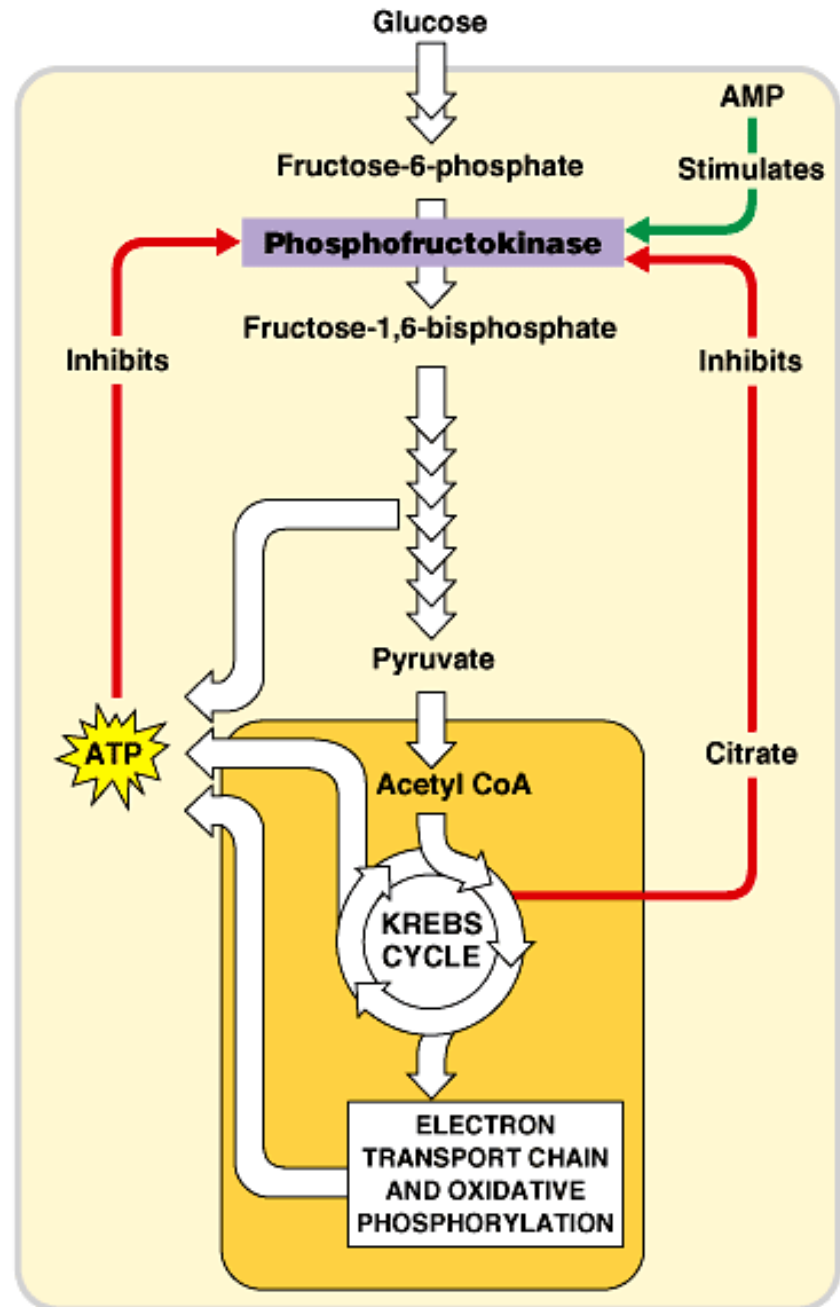
Nucleotide Metabolism

- The many stops on the GATC Line



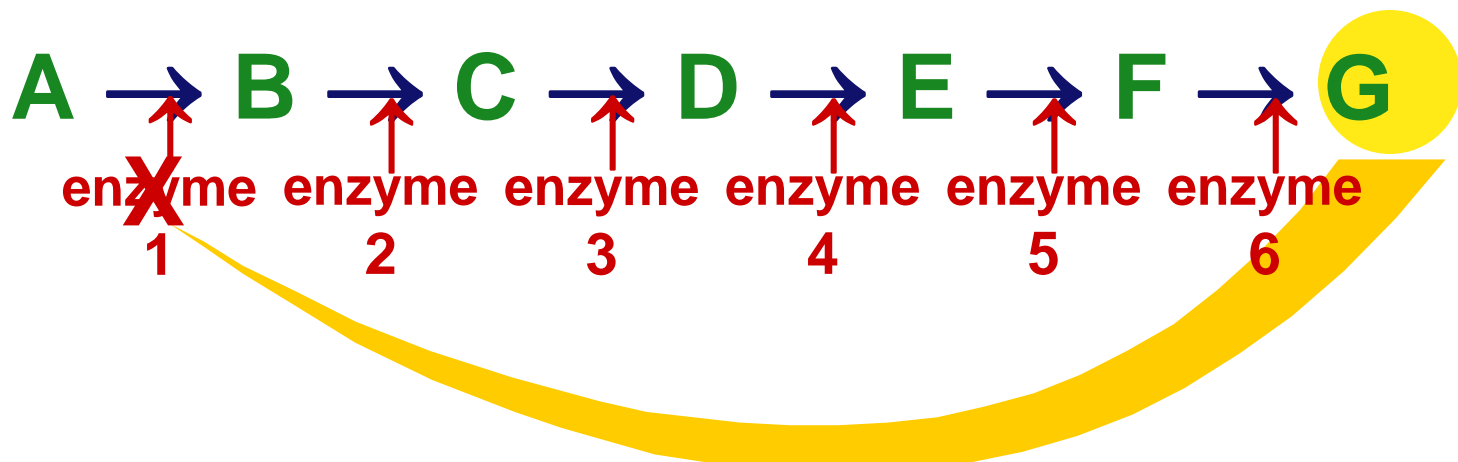
Control of Respiration

Feedback Control



Feedback Inhibition

- Regulation & coordination of production
 - ◆ production is self-limiting
 - ◆ final product is inhibitor of earlier step
 - allosteric inhibitor of earlier enzyme
 - ◆ no unnecessary accumulation of product

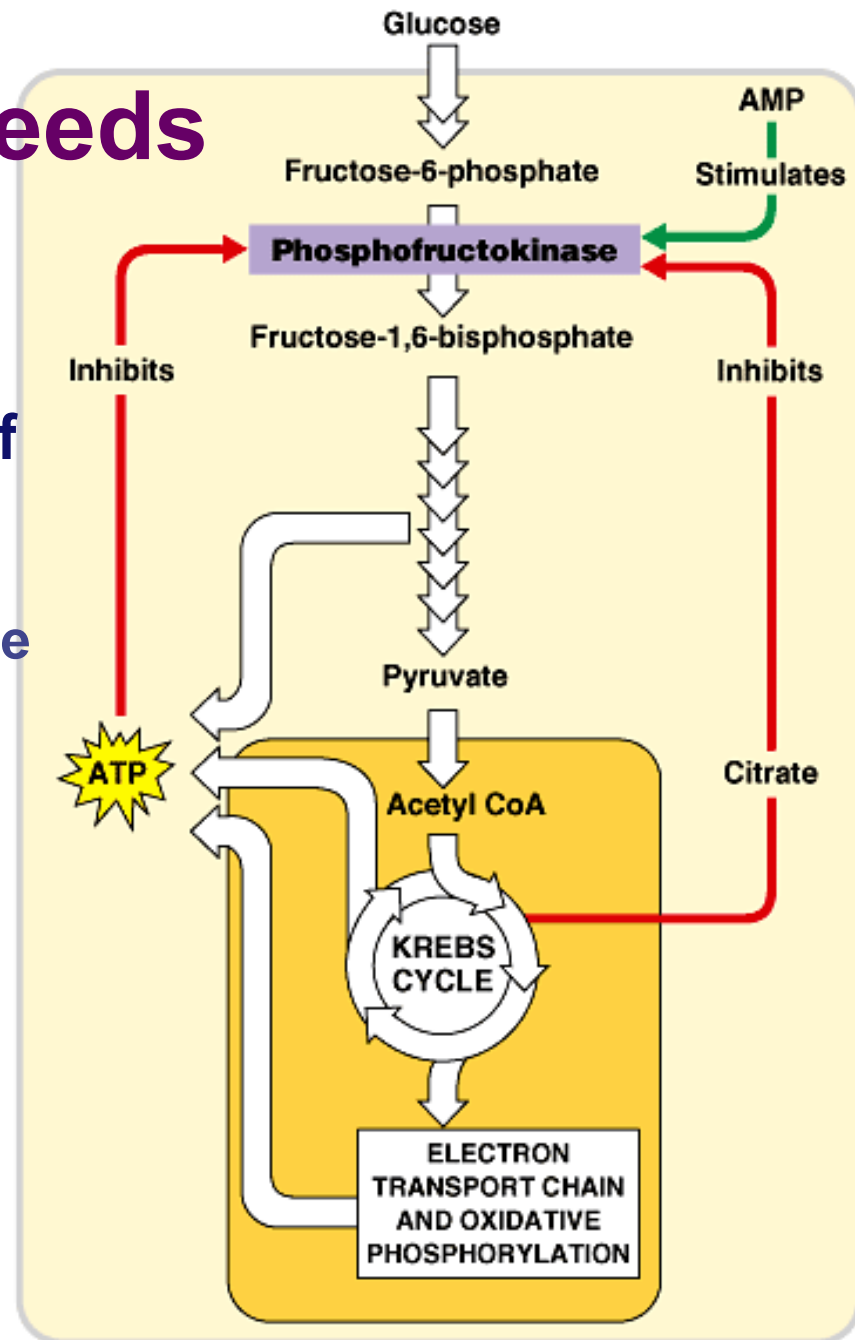


Respond to cell's needs

- Key points of control
 - ◆ phosphofructokinase
 - allosteric regulation of enzyme
 - ◆ “can't turn back” step before splitting glucose
 - AMP & ADP stimulate
 - ATP inhibits
 - citrate inhibits

Why is this regulation important?

Balancing act:
availability of raw materials vs.
energy demands vs. synthesis



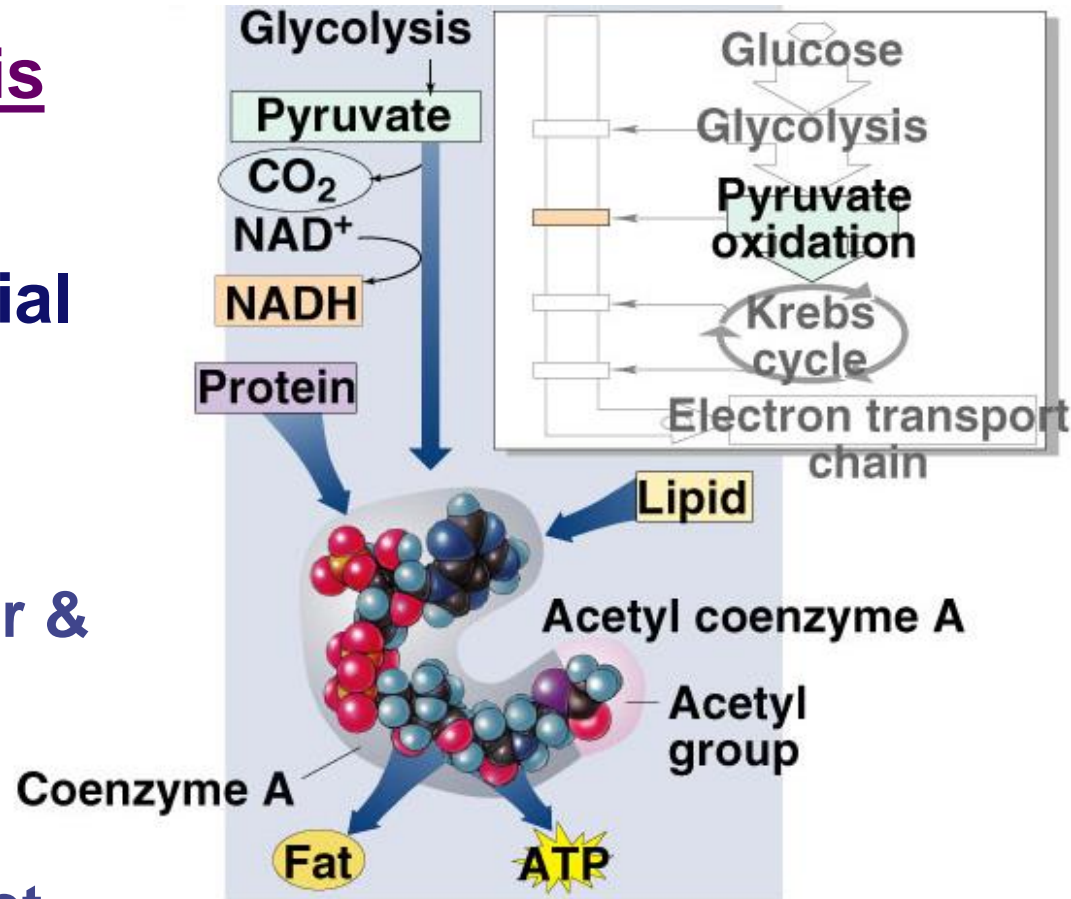
A Metabolic economy

- **Basic principles of supply & demand regulate metabolic economy**
 - ◆ balance the supply of raw materials with the products produced
 - ◆ these molecules become feedback regulators
 - they control enzymes at strategic points in glycolysis & Krebs cycle
 - ◆ AMP, ADP, ATP
 - regulation by final products & raw materials
 - ◆ levels of intermediates compounds in the pathways
 - regulation of earlier steps in pathways
 - ◆ levels of other biomolecules in body
 - regulates rate of siphoning off to synthesis pathways

It's a Balancing Act

- Balancing synthesis with availability of both energy & raw materials is essential for survival!

- ◆ do it well & you survive longer
- ◆ you survive longer & you have more offspring
- ◆ you have more offspring & you get to “take over the world”



Acetyl CoA is central to both energy production & synthesis
make ATP or store it as fat

A decorative graphic consisting of a horizontal blue line extending from the left edge to the right, and a vertical blue line extending from the top edge to the bottom. At the top-left corner, there is a small white circle with a blue outline. At the bottom-right corner, there is a similar small white circle with a blue outline.

Any Questions??